

#### Maharogi Sewa Samiti, Warora

### ANAND NIKETAN COLLEGE, ANANDWAN



# **BOTANICAL SOCIETY**

**ORGANIZING** 

# BOTANY EXAMTHON-2021

Online MCQ based series of Unit Tests and Prelims in the subject Botany for PG students of Gondwana University, Gadchiroli from 01<sup>st</sup> August to 12<sup>th</sup> August, 2021.

#### **OBJECTIVES**

- **❖** To revise topic contents in detail with reasoning ability.
- **❖** To practice solving of MCQs in prescribed duration.
- **❖** To bring attention of student and build subject interest.
- To make students attentive in online examination patterns.
- To build confidence among students and restrain them from indulging in unfair activities during University Online Examinations.

#### **FEATURES OF THE ACTIVITY**

- ✓ Test examination on each prescribed Unit of the Theory Paper.
- ✓ Prelims on the entire Theory Paper prescribed in the syllabus.
- ✓ Answer Feedback with reference texts, diagrams, videos and links.
- ✓ On request repeat examination facility.

#### **ELIGIBILITY AND RULES FOR PARTICIPATION**

- Regular and External M. Sc. Botany students of even semesters (II and IV) studying in Colleges affiliated to Gondwana University, Gadchiroli are eligible for participation.
- Student has to download Google Classroom App in the device.
- > Student has to join Google Classroom using the code o4akq5w or by clicking the link <a href="https://classroom.google.com/c/Mzc1MTI5NzEwMDAw?cjc=o4akq5w">https://classroom.google.com/c/Mzc1MTI5NzEwMDAw?cjc=o4akq5w</a>
- Links for the scheduled examination will be shared on the Google Classroom only.
- > Any query and request will be entertained through message only to 8766916089.
- ➤ Student participants attending all Unit Tests and Prelims and exhibiting consistent performance by scoring 40 % and above marks will receive e-Certificate with following Botanical Society Badges and Titles;
  - Outstanding Performance scoring 80% and above,
  - **Excellent Performance scoring 60 to 79 %.**

### Tentative Time Table for Online MCQ based Unit Test BOTANY EXAMTHON, Summer - 2021

Date	Day	Time	Class/Semester	Paper (Unit)
		9:00 to 9:30 am	M. Sc. SEM II	Paper V: Plant Physiology & Biochemistry; Unit I: Photosynthesis
			M. Sc. SEM IV	Paper XIII: Plant Biotechnology; Unit I: RDT. Genetic Engineering of Plants
		12:00 to 12:30 pm	M. Sc. SEM II	Paper V: Plant Physiology & Biochemistry; Unit II: Respiration
04 /00 /2024	Consider		M. Sc. SEM IV	Paper XIII: Plant Biotechnology; Unit II: Microbial Genetic manipulation, Genomics, Proteomics
01/08/2021	Sunday	6:00 to 6:30 pm	M. Sc. SEM II	Paper V: Plant Physiology & Biochemistry; Unit III: Metabolisms and Nutrient Cycles
			M. Sc. SEM IV	Paper XIII: Plant Biotechnology; Unit III: PTC, Transgenic Production
01/08/2021 Si		9,00 to 9,30 mm	M. Sc. SEM II	Paper V: Plant Physiology & Biochemistry; Unit IV: Enzymology and Solute Transportation
		8:00 to 8:30 pm	M. Sc. SEM IV	Paper XIII: Plant Biotechnology; Unit IV: Bioinformatics
	ă.	0.00 to 0.20 am	M. Sc. SEM II	Paper VI: Plant Development and Reproduction; Unit I: Shoot Development
	<i>/</i> 07	9:00 to 9:30 am	M. Sc. SEM IV	Paper XIV: Angiosperms-II; Unit I: Magnolidae, Hamamelidae, Dilleniidae, Rosidae, Asteridae
	<i>(i)</i>		M. Sc. SEM II	Paper VI: Plant Development and Reproduction; Unit II: Leaf, Root, Flower Development
	)	12:00 to 12:30 pm	M. Sc. SEM IV	Paper XIV: Angiosperms-II; Unit II: Alismatidae, commelinidae, Aracidae, Lilidae, Cucurbitaceae, Cactaceae, Asteraceae, Amentiferae, Lemnaceae, Palmae, Orchidaceae
02/08/2021	Monday	6:00 to 6:30 pm	M. Sc. SEM II	Paper VI: Plant Development and Reproduction; Unit III: Male, Female Gametophyte Development, Pollen-Pistil Interactions
			M. Sc. SEM IV	Paper XIV: Angiosperms-II; Unit III: Probable ancestors of angiosperms, primitive living angiosperms, speciation, extinction, IUCN categories, distribution & global pattern of biodiversity.
		8:00 to 8: <mark>30 pm</mark>	M. Sc. SEM II	Paper VI: Plant Development and Reproduction; Unit IV: Seed & Fruit Development, PCD
			M. Sc. SEM IV	Paper XIV: Angiosperms-II; Unit IV: Biodiversity
	T	9:00 to 9:3 <mark>0 am</mark>	M. Sc. SEM II	Paper VII: Cell & Molecular Biology-I; Unit I: Cell Wall, Plasma Membrane, Plasmodesmata
			M. Sc. SEM IV	Paper XV: Molecular Biology and Plant Biotechnology-II; Unit I: Transgenics
		12:00 to 12:30 pm	M. Sc. SEM II	Paper VII: Cell and Molecular Biology-I; Unit II: Cellular Organelles, Cell Shape and Motility
02/08/2024			M. Sc. SEM IV	Paper XV: Molecular Biology and Plant Biotechnology-II; Unit II: Transformation
03/08/2021	Tuesday	C.00 to C.20 pm	M. Sc. SEM II	Paper VII: Cell and Molecular Biology-I; Unit III: Nucleus
	1	6: <mark>00 to 6:30 pm</mark>	M. Sc. SEM IV	Paper XV: Molecular Biology and Plant Biotechnology-II; Unit III: PTC
		9,00 to 9,30 pm	M. Sc. SEM II	Paper VII: Cell and Molecular Biology-I; Unit IV: Molecular Biology of Stress Response
		8:00 to 8:30 pm	M. Sc. SEM IV	Paper XV: Molecular Biology and Plant Biotechnology-II; Unit IV: Cleaner Biotechnology
		9:00 to 9:30 am	M. Sc. SEM II	Paper VIII: Angiosperm-I; Unit I: Angiosperm Morphology
		9.00 to 9.50 and	M. Sc. SEM IV	Paper XVI: Basic Botany-II; Unit I: Plant Physiology
		12:00 to 12:20 nm	M. Sc. SEM II	Paper VIII: Angiosperm-I; Unit II: Angiosperm Taxonomy
04/09/2021	Wednesday	12:00 to 12:30 pm	M. Sc. SEM IV	Paper XVI: Basic Botany-II; Unit II: Ecology and Evolution
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		9:00 to 9:30 nm	M. Sc. SEM II	Paper VIII: Angiosperm-I; Unit IV: Biosystematics, Plant Nomenclature
		8:00 to 8:30 pm	M. Sc. SEM IV	Paper XVI: Basic Botany-II; Unit IV: Paleobotany

#### Tentative Time Table for Online MCQ based Pre-Term BOTANY EXAMTHON, Summer – 2021

#### **FIRST ROUND**

Date	Day	Time	Class	Paper
05 /00 /2024	Thursday	7:00 to 9:15 nm	M. Sc. SEM II	Paper V PSCBOTT05 Plant Physiology and Biochemistry
05/08/2021	inursday	7:00 to 8:15 pm	M. Sc. SEM IV	Paper XIII PSCBOTT13 Plant Biotechnology
06/00/2024	Friday	7:00 to 8:15 pm	M. Sc. SEM II	Paper VI PSCBOTT06 Plant Development and Reproduction
06/08/2021			M. Sc. SEM IV	Paper XIV PSCBOTT14 Angiosperms-II
07/08/2021	Saturday	7:00 to 0:15 pm	M. Sc. SEM II	Paper VII PSCBOTT07 Cell and Molecular Biology
		7:00 to 8:15 pm	M. Sc. SEM IV	Paper XV PSCBOTT15 Molecular Biology and Plant Biotechnology
08/08/2021	Consider	Sunday 7:00 to 8:15 pm	M. Sc. SEM II	Paper VIII PSCBOTT08 Angiosperms-I
	Sunday		M. Sc <mark>. SEM IV</mark>	Paper XVI PSCBOTT16 Basic Botany-II

#### SECOND ROUND

Date	Day	Time	Class	Paper
09/08/2021	Monday	7:00 to 8:15 pm	M. Sc. SEM II	Paper V PSCBOTT05 Plant Physiology and Biochemistry
09/08/2021			M. Sc. SEM IV	Paper XIII PSCBOTT13 Plant Biotechnology
10/00/2021	Tuesday	7:00 to 8:15 pm	M. Sc. SEM II	Paper VI PSCBOTT06 Plant Development and Reproduction
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#### **ORGANIZING COMMITTEE**

#### CONVENER

Dr. Prashant J. Wagh Contact Number: 8766916089



#### **PRINCIPAL**

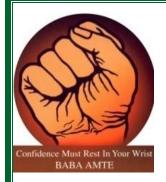
Dr. Mrunal C. Kale
Anand Niketan College, Anandwan



#### COORDINATORS

Ms. Vidya D. Dadmal Mr. Chetan A. Warade





#### Maharogi Sewa Samiti, Warora

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## Gondwana University Gadchiroli B.Sc. Sem VI Online Practical Examination (Mock Test) Summer 2020 Subject: Zoology

Gondwana University Gadchiroli Practical Exam Question-Answer Format

	[As per the University Practical Exam. Scheme]		
	Centre: Anand Niketan College, Anandwan, Warora.		
	Time: 50 Minutes		
	Marks: 30		
*	* Required		
1.	Email *		
2.	Name of Student *		
3.	Roll No. *		
4.	Class *		

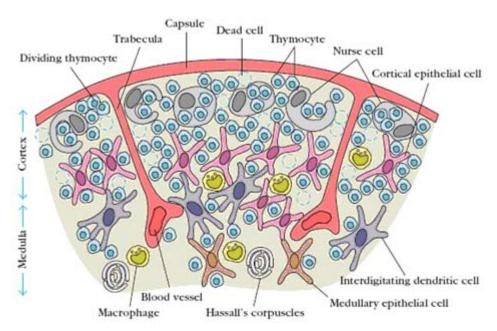
5. PRN No.

6.	Mobile No. *
7.	E-mail ID *
8.	Date of Exam *
	Example: January 7, 2019

Identification of Spot

#### 9. 1. Identify the given spot

1 point

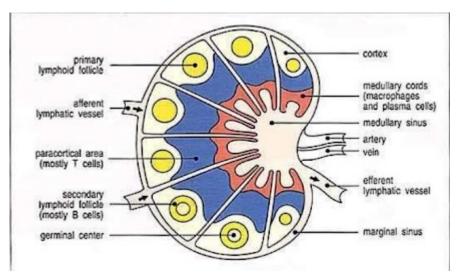


Mark only one oval.

- \_\_\_\_ Lymph Node
- Spleen
- \_\_\_\_ Thymus
- ( Liver

#### 10. 2. Identify the following spot:

1 point

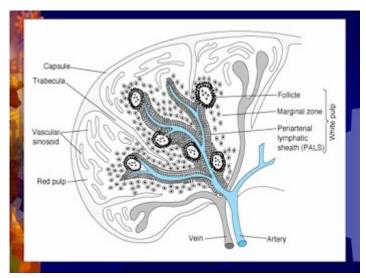


Mark only one oval.

- Spleen
- Lymph Node
- \_\_\_\_ Thymus
- None of the above

#### 11. 3. Identify the given spot;

1 point



Mark only one oval.

	Spl	leen
_	-	

- \_\_\_\_ Lymph Node
- Bone Marrow
- \_\_\_\_ Thymus
- 12. 4. Which of the following cell mature in the Thymus?

1 point

Mark only one oval.

- B-Cell
- T-Cell
- Macrophage
- Dendritic Cell

13.	5. Which of the following is the primary lymphoid organ?	1 point
	Mark only one oval.	
	Lymph Node	
	Spleen	
	Bone Marrow	
	MALT	
Ma	ajor Experiment Immunology	
14.	1. Write a short note on the material required for Ouchterlony Double- Diffusion. *	1 point
15	2 M/s also a service of O ablades De ble Diff air 2 *	
15.	2. Why do we perform Ouchterlony Double-Diffusion? *	1 point

16.	3. What is the antibody excess zone (prozone phenomenon)?	1 point
	Mark only one oval.	
	An increasing amount of antigen added to constant amount of antibody	
	An increasing amount of antibody added to constant amount of Antigen	
	An increasing amount of antibody added to increasing amount of Antigen	
	Constant amount of Antibody added to constant amount of antigen	
17.	4. The precipitin lines will be visible in	1 point
	Mark only one oval.	
	1-2 hours	
	4-8 hours	
	10-12 hours	
	24-48 hours	
Mi	nor Experiment Immunology	
18.	1. What is the location of the antigen in RBC?	1 point
	Mark only one oval.	
	Cytoplasm	
	Nucleus	
	Cell Surface	
	Mitochondria	

19.	2. If there is no clumping in all three cavities, then the blood sample contains blood group	point
	Mark only one oval.	
	AB +ve	
	O -ve	
	O +ve	
	B -ve	
Se	ection Cutting	
20.	1. The thickness of the sections should be	point
	Mark only one oval.	
	0.05 micron	
	0.5 micron	
	5.0 micron	
	50 micron	
21.	2. What causes sections curl into tubes as they are cut and ribbon not form 1	point
	as a result?	
	Mark only one oval.	
	The sections do tend to curl in the beginning	
	Knife angle inclined too much	
	Knife is not sufficiently sharp	
	All of the above	
St	aining	

22.	1. What is the use of xylene?	1 point
	Mark only one oval.	
	Clearing agent	
	Stain	
	Dehydrating agent	
	Mounting material	
23.	2. Which of the following stains are used for the double staining method?	1 point
	Mark only one oval.	
	Safranin - Hematoxylene	
	Hematoxylene - Methylene Blue	
	Eosin - Hematoxylene	
	Eosin - Methylene Blue	
24.	3. Choose the correct staining sequence	1 point
	Mark only one oval.	
	Deparaffinization - Hydration - Staining - Washing - Dehydration	
	Washing - Dehydration - Staining - Hydration - Deparaffinization	
	Deparaffinization - Dehydration - Washing - Staining - Hydration	
	Staining - Dehydration - Washing - Deparaffinization - Hydration	
Bio	oinformatics	

25.	1. Which of the following is NOT a protein sequence database?	1 point
	Mark only one oval.	
	PIR	
	SWISS-PROT	
	GenBank	
	UniProt	
26.	2. Which of the following tool is used for sequence alignment?	1 point
	Mark only one oval.	
	BLAST	
	FASTA	
	Pfam	
	All of the above	
Bio	ostatistics	
27.	1. What is the mode of the following distribution? 25, 17, 23, 23, 24, 25, 23	1 point
	Mark only one oval.	
	25	
	23	
	17	
	24	

28.	2. Find the mean following distribution? 12, 15, 12, 14, 15, 18	1 point
	Mark only one oval.	
	14.3	
	14	
	13.5	
	12	
29.	3. Which of the following is a graphic representation of a cumulative frequency distribution?	1 point
	Mark only one oval.	
	Frequency Polygon	
	Pie Charts	
	Line Chart	
	Ogive	
۷IV	VA	
30.	1. Which microtome did you use for section cutting?	1 point
	Mark only one oval.	
	Rocking	
	Rotary	
	Locomotory	
	Rockorotary	

31.	2. Which of the following is NOT a clearing agent?	1 point
	Mark only one oval.	
	Xylene	
	Cedar Wood Oil	
	Alcohol	
	Clove Oil	
32.	3. Acquired immunity is known as	1 point
	Mark only one oval.	
	Specific immunity	
	Non-specific immunity	
	Innate immunity	
	Primary line of defense	
33.	5. Which of the following Immunoglobulin is present in Pentameric form	1 point
	Mark only one oval.	
	☐ IgG	
	☐ IgD	
	☐ IgM	
	All of the above	
D۳	ractical Record	
FI	actical Necolu	

34.	Have you submitted a certified practical record?	5 points
	Mark only one oval.	
	Yes	
	No	

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## Google Forms

# Gondwana University Gadchiroli B.Sc. Sem VI Online Practical Examination (Mock Test) Summer 2020 Subject: Zoology

69 responses

Name of Student

69 responses

Vijeta Vasantarao Darokar

Shital devidas wandhare

Ravi Chandrabhan Apate

Shruti Neware

Pratiksha Ashok Narawade

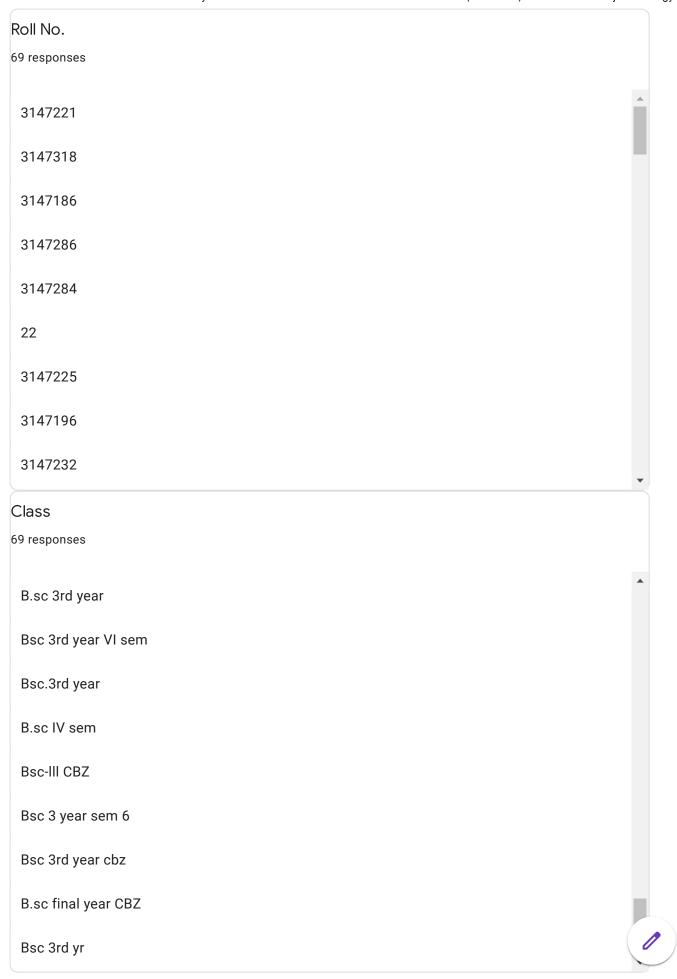
Shraddha dhote

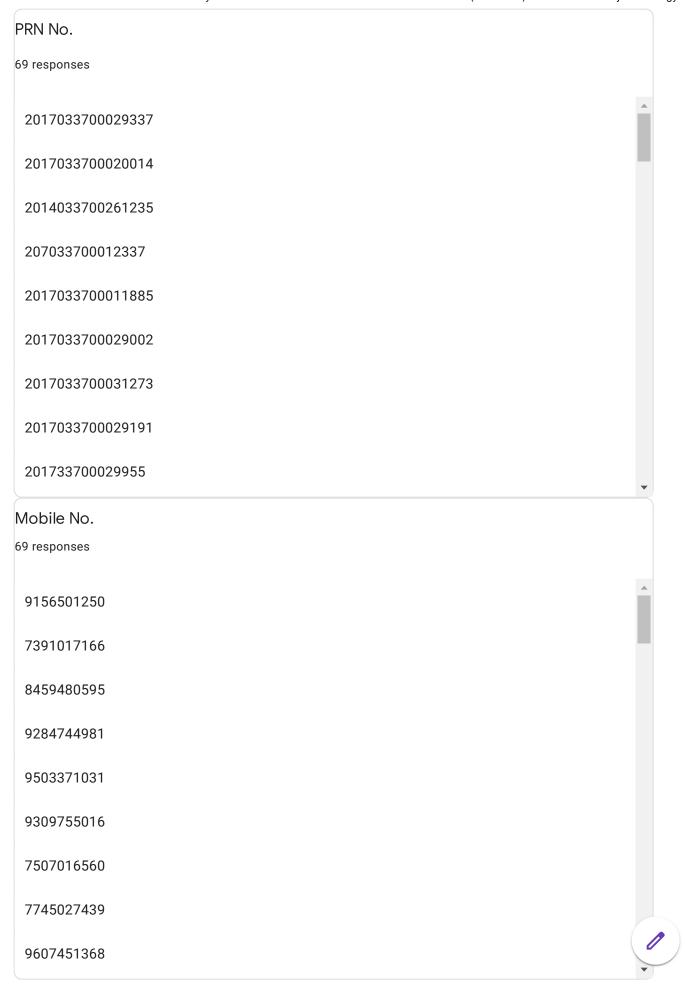
Monali waman dhawas

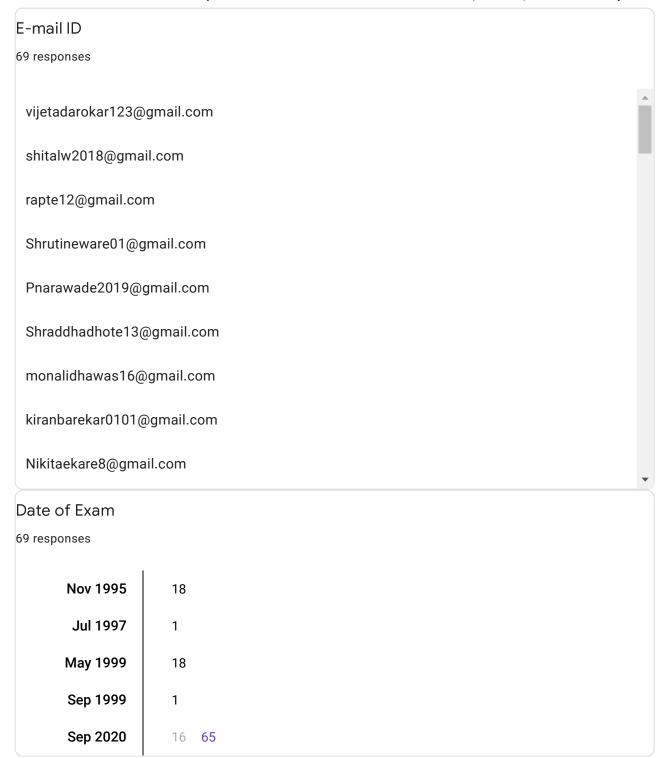
Kiran Sukhadeo Barekar

Nikita bhojraj Ekare



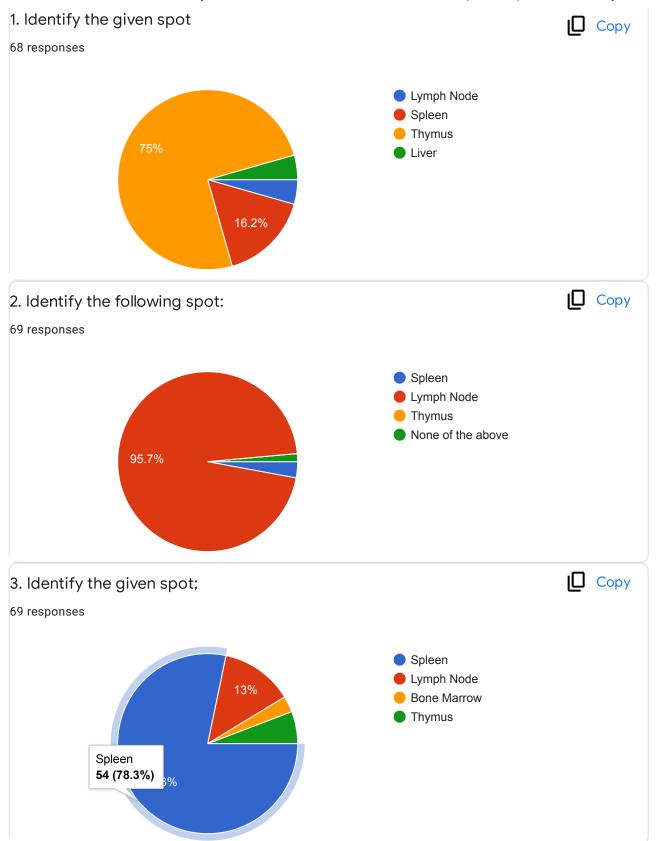




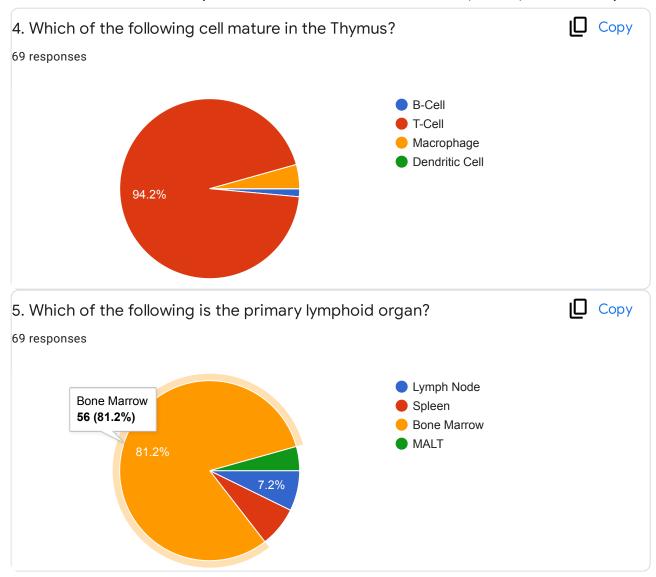


Identification of Spot









Major Experiment Immunology



1. Write a short note on the material required for Ouchterlony Double-Diffusion.

67 responses

Which are useful for the analysis of antigen and antibody.

The double diffusion assay it is frequently useful of comparing different antigen preparation.

Micropipette, microslide, antigen and antibody solution

Micropippette, microslide, antigen solution antibody well puncture

Micropepet, Microslide, antigen solution, antigen solution

Micro pepette, micro slide, antigen solution, well puncture

Immuno diffusion is get in compound the variety of techniques which are useful for the analysis of antigen and antibody complex.double diffusion it is frequently useful in comparing different antigen preparation

Micro pepitte, antigen solution, antiserum, assay buffer, gel punch with syringe, agarose, petri plate distilled water, etc.

2. Why do we perform Ouchterlony Double-Diffusion?

65 responses

To analysis antigen and antibody

No

Demonstration how to occur the degree of similarity of antigen.

Different antigen preparation each containing the single antigenic species.

Similarity between antigen and antiserum well pattern of line. It can be illustrated. (1) pattern of identity. (2) pattern of partial identity (3) pattern of non identity

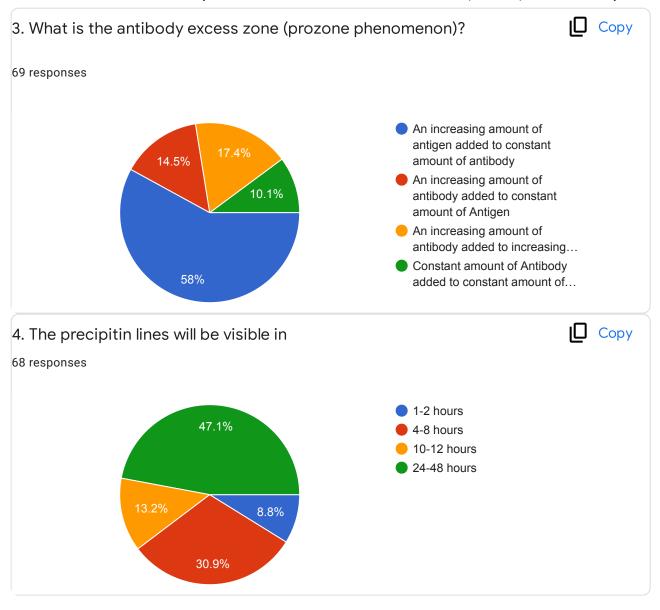
Agarose gel and prepare the microslide .antigen and antibody solution 25 to 100 microliter

Antigen and antibody are allow to diffuse into the gel.

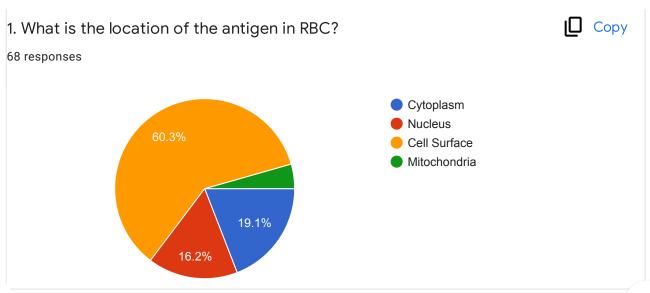
The ouchterlony double diffusion technique is one of the simplest techniques extensively used to check antisera for the presence of antibodies for a particular Ag and to determine Titre.

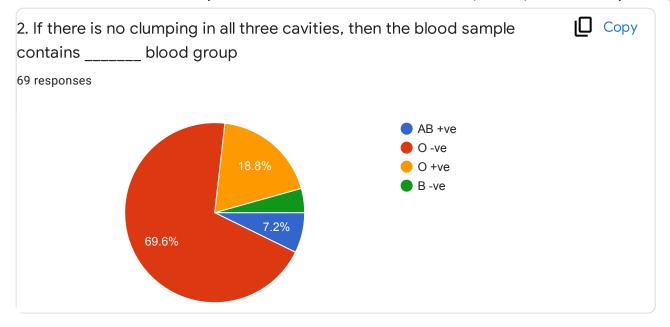
Because this technique useful for the analysis of antigen and antibody



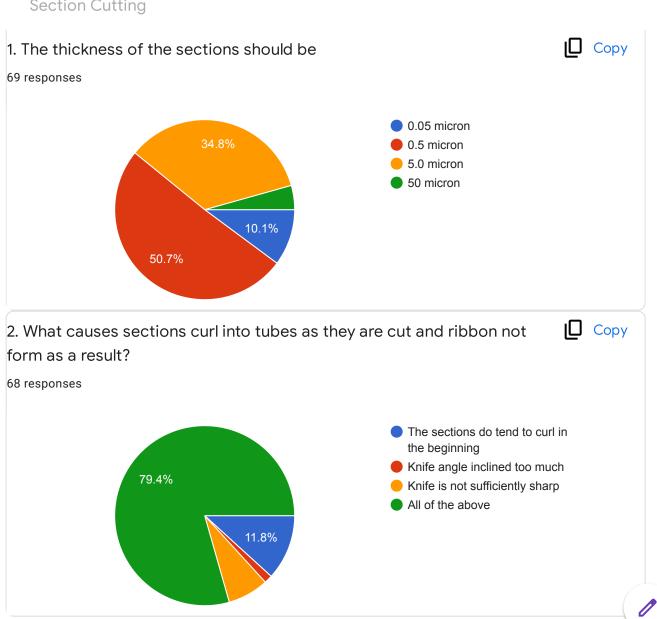


Minor Experiment Immunology

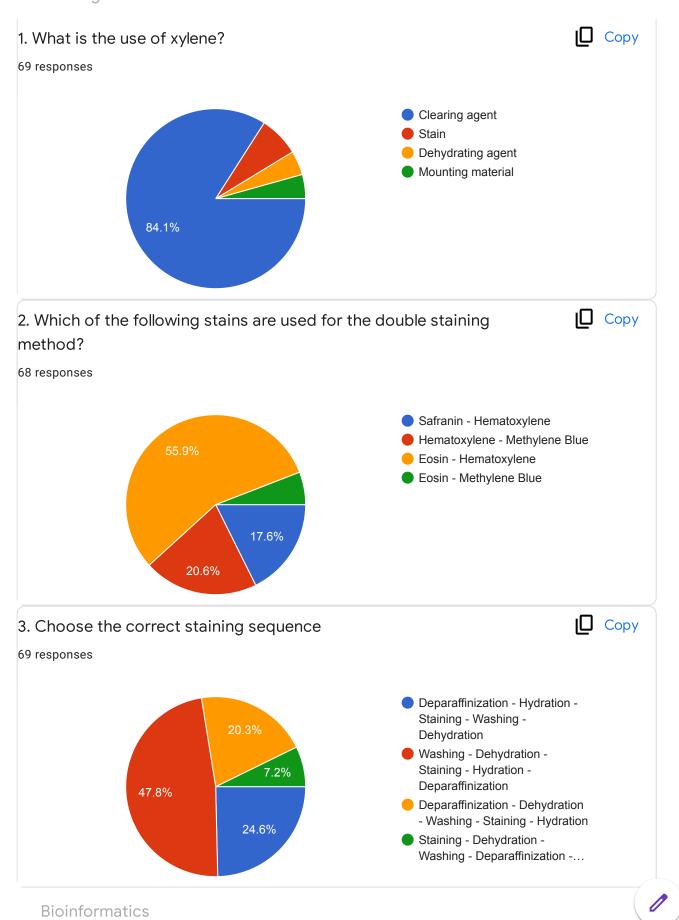


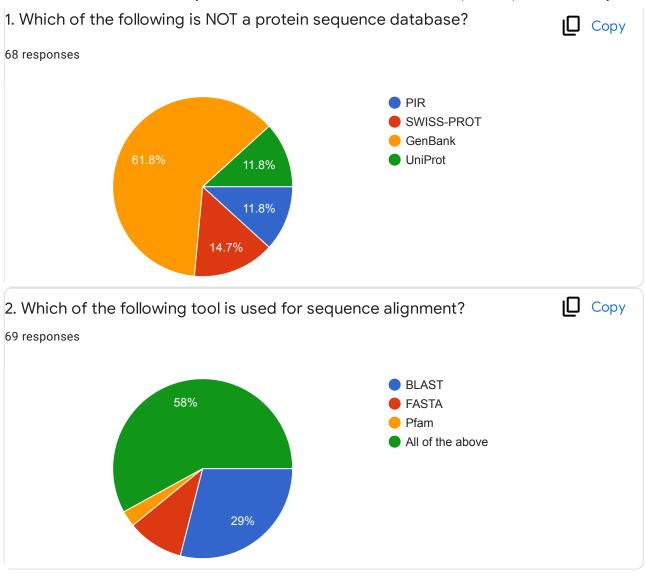


**Section Cutting** 

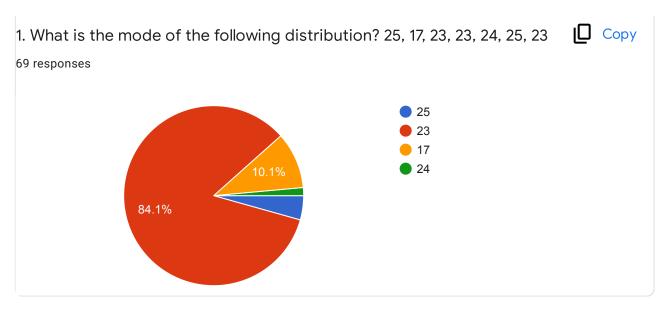


Staining

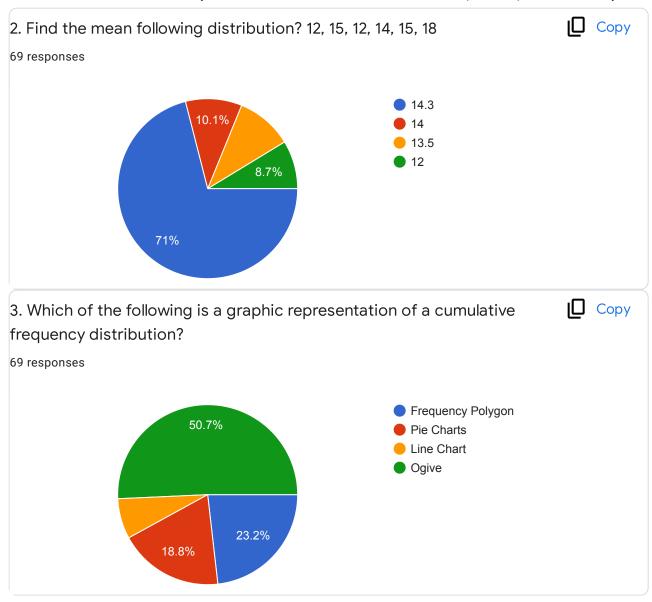




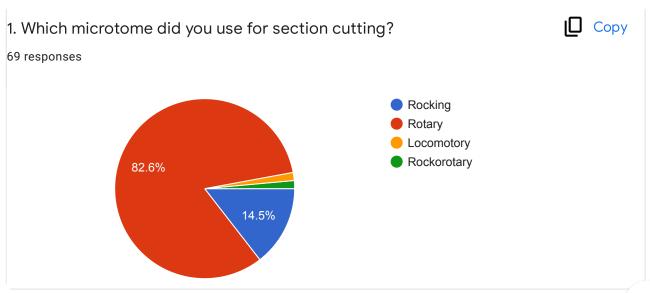
**Biostatistics** 

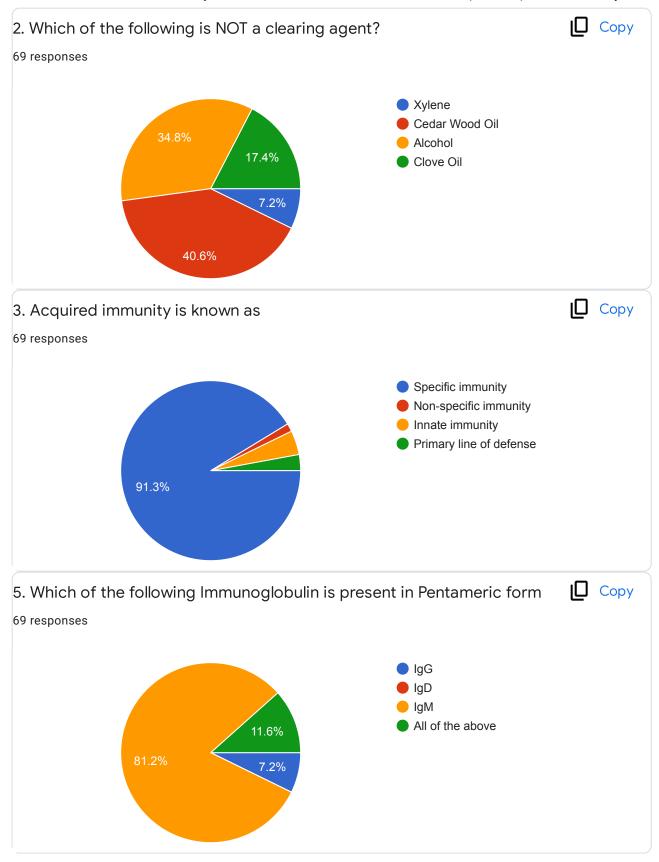






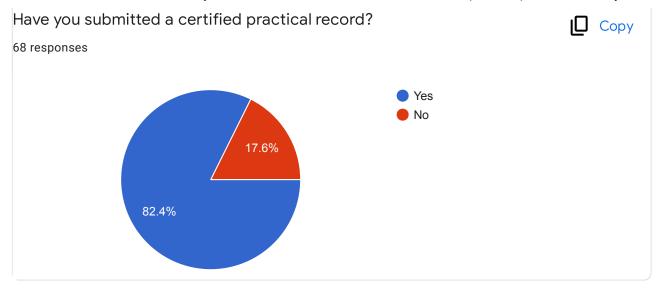
VIVA





Practical Record





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## MOCK TEST { Practical Exam - 2021}

BSc. Part -2 Sem - 4
Subject - Chemistry Practical
Organized by
Department Of Chemistry
Anand Niketan College, Warora

Note -1] All questions are compulsory. 2] Each questions carry one Marks.

*	Required		
1.	Email *		
2.	Name Of The Student *		
0			
3.	Class - BSc. 2nd Year [PCM / CBZ ] *		
	Mark only one oval.		
	PCM		
	CBZ		
4.	Mobile Number *		
Q	Que. No1		

5.	1) Name the reagent used to detect carbonyl group in a compound *
	Mark only one oval.
	NaHCO3
	Carbylamine
	2,4 dinitrophenyl hydrazine
	None of the above
Ç	Que. No2
6.	2) When organic compound is added with saturated NaHCO3 and Brisk effervescence is obtained *
	Mark only one oval.
	Presence of - CH0 group
	Presence of -C00H group
	Presense of carbohydrate
	Presence of amine group
Ç	Que. No3
7.	3) Brown ring test appears in the test for *
	Mark only one oval.
	Nitrate
	Iron
	Nitrite
	Bromide
Ç	Que. No 4

8.	4) The saturated carbon compound burnt completely and gives clear blue flame This is known as
	Mark only one oval.
	Sooty Flame
	Non sooty Flame
	Both
	None of the above
ζ	Que. No5
9.	5) Touch the moist litmus paper with an organic compound & blue litmus turn red indicates -
	Mark only one oval.
	It may be amine
	absence of carboxylic acid & phenol
	absence of amine
	It may be carboxylic acid & Phenol
ζ	Que. No6
10.	6) Lassaignes Test is used in detection of*
	Mark only one oval.
	Nitrogen
	Sulphur
	Halogens
	All of the above
C	Que. No7
~	

11.	7) Na is kept immersed in*
	Mark only one oval.
	Alkane Acetone Kerosene Ether
Qu	e. No8
12.	8) Schiffs reagent is used to detect*
	Mark only one oval.
	aldehyde group
	amine group
	Carboxylic group
	Nitro group
Qu	e. No9
13.	9) In dye test , Formation of Orange azo dye indicates*
	Mark only one oval.
	Presence of aromatic amine
	Presence of aliphatic amine
	Both
	None of the above
Qu	e. No10

14.	10) Tollense Reagent is a *
	Mark only one oval.
	Na / K tartarate
	P-rosaniline hydrochloride
	Ammonical silver nitrate
	CuS04
Qι	ie. Noii
15.	11) When solution of organic compound added with bromine water & Brown colour of
10.	bromine is discharged *
	Mark only one oval.
	Unsaturation is present
	No unsaturation is present
	Compound is aromatic
	Compound is aliphatic
Qι	1e. No12
16.	12) Most of the nitro compounds usually in colour. *
	Mark only one oval.
	Red
	Yellow
	Violet
	Green
Oı.	ie. No13
(-	

1/.	13) Fehling solution A is a
	Mark only one oval.
	aqueous CuSO4
	Na or K tartarate
	ammonical silver nitrate
	P- Rosaniline hydrochloride
Qι	ne. No14
18.	14) The hybridisation of Hexamine Nickel chloride complex is *
	Mark only one oval.
	sp2 Hybridisation
	sp3 Hybridisation
	dsp2 Hybridisation
	sp3 d2 Hybridisation
	NI.
Qt	ie. No15
19.	15) The gravimetric analysis is a*
	Mark only one oval.
	quantitative measurment
	qualitative measurment
	Both
	Neither qualitative nor quantitative
<u>O</u> 1:	ue. No16
Q.	

20.	16) Glucose gives Precipitate with fehling solution. *			
	Mark only one oval.			
	Blue			
	White			
	Reddish brown			
	Yellow			
Qu	e. No17			
21.	17) What is the colour of tetramine copper sulphate complex ? *			
	Mark only one oval.			
	Blue			
	Yellow			
	Red			
	Brown			
Qu	e. No18			
22.	18) Which Method is used to determine composition of Fe-SSA complex. *			
	Mark only one oval.			
	Solubility method			
	Pycnometric method			
	Job's method			
	Viscometric method			
Qu	e. No19			

23.	19)What is the ratio of barium to sulphate ion in BaSO4 *			
	Mark only one oval.			
	1:1			
	1:2			
	2:1			
	1:4			
Qı	ie. No20			
24.	20] Dimethyl Glyoxime is a specific reagent for *			
	Mark only one oval.			
	Fe (+3)			
	Co (+2)			
	Ni (+2)			
	Cu (+2)			
Ou	ya Najar			
Qι	ie.No21			
25.	21) The molecular formula of tetramine copper sulphate complex is *			
	Mark only one oval.			
	[Cu (NH3)4] S04 ,H20			
	[Co (NH3)4] S04 ,H20			
	[Cu(NO3)4] S04 ,H20			
	[Cu(NH4)3] S04 ,H20			
Qt	ie. No22			

26.	22) In estimation of acetamide , NaOH is used as*		
	Mark only one oval.		
	Oxidising agent		
	Reducing agent		
	Hydrating agent		
	None of the above		
Qu	ie. No23		
07	) ml		
27.	23) The Oxidation state of copper in tetramine copper sulphate complex is*		
	Mark only one oval.		
	1		
	+1		
	2		
	+2		
Qu	ne. No24		
28.	24) Dimethyl Glyoxime is a ligand *		
	Mark only one oval.		
	Monodentate		
	Bidentate		
	Tetradentate		
	polydentate		
Viv	va Que1		

29.	1) The aldehyde functional group is *
	Mark only one oval.
	COOH
	- COOR
	CHO
	-0Н
Viv	va Que2
20	\ml 1 1 C 1 C 1 ·
30.	2) The molecular formula of acetamide is *
	Mark only one oval.
	CH3-NH2
	CH3-C0NH2
	СН3-СНО
	CH3-C00-CH3
Viv	va Que3
01	) ml
31.	3) The Formula of % yield =*
	Mark only one oval.
	Actual yield / Theo. yield X100
	Theo.yield / Actual yield X100
	Actual yield X Theo.yield X100
	None of the above

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## Mock test B.Sc Sem-VI

Department of Chemistry.
Anand Niketan College Warora

* F	Required	
1.	Email *	
St	udent Information	
2.	Name of Student *	
3.	Class [ B.Sc Sem -VI] *	
	Mark only one oval.	
	PCM	
	CBZ	
4.	Contact no. *	
Ind	organic chemistry	12 questions 1 mark each

5.	Paper chromatography is a *	1 point
	Mark only one oval.	
	Gas-Gas chromatography	
	Gas-liquid chromatography	
	Solid -liquid chromatography	
	Liquid-liquid chromatography	
6.	The distance travelled by the component of the mixture to the distance travelled by solvent front is called *	1 point
	Mark only one oval.	
	Absorption coefficient	
	Retardation factor	
	Transmission	
	Reflection	
7	Acidia mahama ang kamandishad forma alla sala kaharan dha mana *	4
7.	Acidic nature can be predicted from pH scale between the range *	1 point
	Mark only one oval.	
	7-12	
	4-10	
	1-6	
	1-14	

8.	Phenolpthalein indicator gives ending point changing colour from *	1 point
	Mark only one oval.	
	Colourless to pink	
	Pink to colorless	
	Red to blue	
	Blue to red	
9.	In double titration method for estimation of carbonates and bicarbonates on soil sample, the indicators used are? *	1 point
	Mark only one oval.	
	Methyl orange and methyl red	
	Phenolpthalein and methyl red	
	Phenolpthalein and methyl organge	
	None of them	
10.	Which of the following is correct formula for carbonate and bicarbonate respectively *	1 point
	Mark only one oval.	
	C02 and C03	
	HC03 and C03	
	C03 and HC03	
	C03 and C02	

4/27/22, 11:02 AM Mock test B.Sc Sem-VI

11.	In paper chromatography, the stationary phase is? *	1 point
	Mark only one oval.	
	Solvent in the beaker	
	Water trapped in cellulose fibre of paper	
	Paper it self	
	None of them	
12.	Water is a *	1 point
	Mark only one oval.	
	Polar solvent	
	Non polar solvent	
	Organic solvent	
	Volatile solvent	
13.	Ion-exchange chromatography is used for the separation of *	1 point
	Mark only one oval.	
	O Polar molecules	
	Non polar molecules	
	Both of above	
	None of above	

14.	On addition of methy orange indicator in double titration method, the color changes to *	or of the solution	1 point
	Mark only one oval.		
	Pink		
	Yellow		
	Red		
	Green		
15.	The general range of pH for alkaline soil is *		1 point
	Mark only one oval.		
	Below 3		
	7		
	Above 12		
	3-10		
16.	The measurement of hydroxide ions in given solution is given by *		1 point
	Mark only one oval.		
	pH		
	рОН		
	pKa		
	pKb		
Phy	sical Chemistry	12 questions 1 mark	each

17.	The measure of the capacity of a substance to absorb light of a specified wavelength is equal to the logarithm of the reciprocal transmittance is called *	1 point
	Mark only one oval.	
	Absorbance	
	Transmittance	
	Reflection	
	Diffraction	
18.	What is <b>1</b> max? *	1 point
	Mark only one oval.	
	Wavelength at which the substance absorbs maximum light	
	Wavelength at which the substance absorbs minimum light	
	Wavelength that shows highest intensity	
	Wavelength that shows lowest intensity	
10		
19.	In Beer's Lambert's law, the plot between absorbance and concentration gives *	1 point
	Mark only one oval.	
	Parabola	
	Straight line	
	Sigmoidal curve	
	Hyperbola	

20.	The ratio of sin of the angle of incident to the angle of refraction is called? *	1 point
	Mark only one oval.	
	Retardation factor	
	Absorptivity	
	Refractive index	
	None of them	
21.	The SI unit of polarizability is? *	1 point
	Mark only one oval.	
	Cm2.v	
	Cm2/V	
	Cm/V2	
	Cm2/V2	
22.	Refractive index is denoted by *	1 point
	Mark only one oval.	
	λ	
	η	
	δ	
	None of them	

23.	The process in which molecules adhere to the surface Is called? *	1 point
	Mark only one oval.	
	Absorption	
	Adsorption	
	Dissolution	
	Evaporation	
24.	The SI unit of surface tension is? *	1 point
	Mark only one oval.	
	◯ N.m	
	◯ N/m	
	N.m2	
	N/m2	
25.	In an experiment to determine surface tension, the instrument used is? *	1 point
	Mark only one oval.	
	Stalagmometer	
	Viscometer	
	pH meter	
	Density bottle	

26.	What is the unit of absorbance? *	1 point
	Mark only one oval.	
	N.m	
	Dyne/cm	
	Cm2	
	Unit less	
27.	What is oxidation state of Mn in KMnO4? *	1 point
	Mark only one oval.	
	+2	
	+6	
	+7	
	+4	
28.	Density can be calculated as *	1 point
	Mark only one oval.	
	◯ M/V	
	◯ V/M	
	C.m	
	◯ M.V	
Viv	a Questions	Each question carries 1 mark
•••		

29.	What is the wavelength range of visual light in electromagnetic spectrum? *	1 point
	Mark only one oval.	
	200-400nm	
	400-700nm	
	700-1200nm	
	2-4 nm	
30.	Which of the following solvents is missile in water?*	1 point
	Mark only one oval.	
	Benzene	
	Ethanol	
	Cyclohexane	
	Napthalene	
31.	The covalent bond becomes polar due to *	1 point
	Mark only one oval.	
	Onization potential	
	Electronegativity difference	
	Heat of hydration	
	Electron affinity	
Tho	ank you	

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Mock Test Chemistry Practical Exam - 2021
BSc. Part -2 Sem - 3
Subject - Chemistry Practical
Organized by
Department Of Chemistry
Anand Niketan College, Warora
Note -1] All questions are compulsory.
2] Each questions carry one Marks.
* Required
1.
Email *
Name Of The Student *
Roll Number *
PRN Number *
Mobile Number *
Email *
Que. No.-1
1] The sodium salt of fatty acid is known as ----- *
Mark only one oval.
Baking Soda
Indicator
Soap
Catalyst
Email *
Que. No.-2
7.
2] The number of molecules taking part in a chemical reaction is known as ----- *
Mark only one oval.
Order of reaction
molecularity
End point
Stability Constant
Email *
Que. No.-3
3] The number of molecules whose concentration determine the rate of chemical reaction at a
given temperature *
Mark only one oval.
Order of reaction
Valency
Molecularity
Stability Constant
Email *
Que. No.- 4
4] The substance which does not involved in a chemical reaction but it increases the rate of
reaction *
Mark only one oval.
```

Redusing agent

oxidising agent

Catalyst

Halogens

Email \*

Que. No.-5

10.

5] The unit of rate constant for first order reaction is ----- \*

Mark only one oval.

mole / L per unit time

1/ time

L/mole per unit time

time /1

Email \*

Que. No.-6

11.

6] Phenol and water are -----Liquid. \*

Mark only one oval.

Completely miscible

Partially miscible

**Immiscible** 

None of the above

Email \*

Que. No.-7

12.

7] The minimum temperature above which the two liquids are completely miscible in all proportion \*

Mark only one oval.

Critical solution temperature

Boiling temperature

**Melting Point** 

Triple point temperature

Email \*

Que. No.-8

13.

8] When the salts like NaCl is added to the reaction mixture. The critical solution temperature is --

Mark only one oval.

Decreases

Increases

Remains Same

None of the above

Email \*

Que. No.-9

14

9] Hydrolysis of methyl acetate gives ----- \*

Mark only one oval.

Acetic acid & ethyl alcohol

Formic acid & ethyl alcohol

Acetic acid & Methyl alcohol

Formic acid & Methyl alcohol

Email \*

Que. No.-10

15.

10] Hydrolysis of ethyl acetate is a ----- reaction \*

Mark only one oval.

Zero order reaction

Pseudo unimolecular first order reaction

Second order reaction

None of the above

Email \*

Que. No.-11

16.

11] If the solute is dissolved in a solvent then the melting point of solvent -----. \*

Mark only one oval.

lowered

Increases

Remains Constant

None of the above

Email \*

Que. No.-12

17.

12] The temperature at which solid and liquid phase exist in equilibrium \*

Mark only one oval.

Boiling point

Melting point

End point

Transition temperature

Email \*

Que. No.-13

18.

13] Which of the following is phase rule?\*

Mark only one oval.

F = C - P + 2

F = C + P + 2

F = P - C + 2

F = P + C + 2

Email \*

Que. No.-14

19.

14] If number of components (C) = 3, number of phases (P) = 1, Then degree of freedom (F) = ?\* Mark only one oval.

6

5

0

4

Email \*

Que. No.-15

20

15] The molecule that contain at least one unpaired electron called as ----- \*

Mark only one oval.

Functional group

Prosthetic group

Free radicals

Semimetals

Email \*

Que. No.-16

21

16] The negatively charged radicals is called as -----radicals \*

Mark only one oval.

acidic

basic

4/27/22, 11:25 AM Mock Test Chemistry Practical Exam - 2021 Neutral radicals None of the above Email \* Que. No.-17 17] When ionic product ----- the solubility product then precipitation takes place . \* Mark only one oval. Equal to greater than Less than Less or equal to Email \* Que. No.-18 23. 18] The quantity of sample required for semi microanalysis is ----- \* Mark only one oval. 0.1 to 5 mg 1 to 10 mg 10 to 100 mg 100 mg to 1 gm Email \* Que. No.-19 24. 19] Due to common ion effect weak electrolyte becomes ----- \* Mark only one oval. More weak More strong remains as it is Either weak or strong Email \* Que. No.-20 20] Dimethyl Glyoxime is a specific reagent for ---- \* Mark only one oval. Fe (+3) Co(+2)Ni (+2)Cu (+2) Email \* Que.No.-21 21] The degree of dissociation of H2S is suppressed by addition of -----strong electrolyte \* Mark only one oval. KC1 HC1 NH4C1 Na2CO3 Email \* Que. No.-22 27. 22] The color of Agl (Silver lodide) \* Mark only one oval. Orange Pink

Blue Yellow Email \*

Que. No.-23

28.

23] Brown ring test is a -----test \*

Mark only one oval.

Sulphides

Nitrate

Halides

Carbonate

Email \*

Que. No.-24

29.

24] The group reagent for halides is ----- \*

Mark only one oval.

Silver Nitrate

Dil HCl

H2S gas in presence of dil. HCl

NH4OH in presence of NH4Cl

Email \*

Viva Que.-1

30.

1] In Laboratory burner ,we use \*

Mark only one oval.

**Producer Gas** 

Oil Gas

Gobar Gas

Coal Gas

Email \*

Viva Que. -2

31.

2] Which of the following is a colligative property \*

Mark only one oval.

Molality

Viscosity

Relative Lowering of vapour pressure

Surface tension

Email \*

Viva Que. -3

32.

3]When common salt is dissolved in water \*

Mark only one oval.

Melting point of solution increases

Boiling point of solution increases

Boiling point of solution decreases

Both melting point and boiling point decreases

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## ANAND NIKETAN COLLEGE, WARORA DEPARTMENT OF PHYSICS, GONDWANA UNIVERSITY EXAMINATION (MOCK TEST) SUMMER-2021.B.Sc. 3rd Year SEM-VI

Name of Examination: Gondwana University SEC Examination Summer-2021(MOCK TEST)

Date:23/07/2021

Centre Name: Anand Niketan College, Anandwan, Warora.

Centre Code: 344

Subject: Basic instrumentation skill

* F	Required	
1.	Email *	
2.	<u>Sr.No.as</u> per list	
3.	Name of Student *	
4.	Seat No./Roll. No. *	
5.	PRN Number *	

6.	Subject *		
7.	Medium *		
8.	Date of Exam *		
	Example: January 7, 2019		
9.	Name of Center *		
A.	Project		Date-23/07/21
10.	1. Digital multimeter is used for	*	2 points
	Mark only one oval.		
	a) measuring a.c. and d.c. current, volta	age and resistance	
	b) measuring a.c. current and voltage		
	c) measuring d.c. current and resistance		
	d) measuring a.c. voltage and resistant	ee	

11.	2.Current is converted to voltage*	2 points
	Mark only one oval.	
	a) through a voltmeter	
	b) through a resistance	
	c) through an ammeter	
	d) through a galvanometer	
12.	3. An ammeter *	2 points
	Mark only one oval.	
	Is inserted in series in a circuit and current to be measured flows through it	
	Is inserted in series in a circuit and part of the current to be measured flows through it	
	Is connected in parallel in a circuit and current to be measured flows through	it
	Is connected in parallel in a circuit and only part of the current to the measure flows through it.	ed
13.	4.For resistance measurement, meter contains *	2 points
	Mark only one oval.	
	a) high current source	
	b) medium current source	
	c) low current source	
	d) low voltage source	

14.	5.Quantities are digitised using *	2 points
	Mark only one oval.	
	a) D/A converter	
	b) oscillator	
	c) amplifier	
	d) A/D converter	
15.	6.Analog mulimeters require power supply. *	2 points
	Mark only one oval.	
	a) True	
	b) False	
16.	7.Output of a digital multimeter is *	2 points
	Mark only one oval.	
	a) mechanical	
	b) optical	
	c) electrical	
	d) analog	
47	8 Rasic building blocks of digital multimator are	
17.	8.Basic building blocks of digital multimeter are *	2 points
	Mark only one oval.	
	a) oscillator, amplifier	
	b) diode, op amp	
	c) rectifier, schmitt trigger	
	d) A/D, attenuator, counter	

18.	9.Resistance is measured using*	2 points
	Mark only one oval.	
	a) constant current source	
	b) constant voltage source	
	c) variable current source	
	d) variable voltage source	
19.	10.A.C. voltages are measured using*	2 points
	Mark only one oval.	
	a) oscillators and op amps	
	b) rectifiers and filters	
	c) resistor and capacitor	
	d) inductor and resistor	
R	Theory	
D.	THEOTY	
20.	1.Low voltage signals can be measured by *	2 points
	Mark only one oval.	
	a) amplifiers	
	b) transformers	
	c) transducers	
	d) voltmeters	

21.	2. Input impedance of an electronic voltmeter is *	2 points
	Mark only one oval.	
	a) low	
	b) high	
	c) medium	
	d) zero	
22.	3.What is basic range of a meter? *	2 points
	Mark only one oval.	
	a) causes maximum deflection	
	b) causes minimum deflection	
	c) causes zero deflection	
	d) causes medium deflection	
23.	4.High range can be obtained in a basic D.C. electronic voltmeter by	2 points
	Mark only one oval.	
	a) a transformer	
	b) an attenuator	
	c) a transducer	
	d) a resistor	
24.	5.Sensitivity of electronic voltmeter is high. *	2 points
		2 pointo
	Mark only one oval.	
	a) True	
	b) False	

25.	6.An ammeter is convertible to a voltmeter by *	2 points
	Mark only one oval.	
	Changing the scale	
	Putting a large resistance in parallel with the actual measuring part of the instrument	
	Putting a large resistance in series with the actual measuring part of the instrument	
	Simply installing the instrument in parallel with the circuit	
26.	7.Overloading is *	2 points
	Mark only one oval.	
	a) damages the meter	
	b) increases the temperature	
	c) doesn't affect the meter	
	d) decreases the sensitivity	
27.	8.Digital voltmeters can be used to measure*	2 points
	Mark only one oval.	
	a)voltage only	
	b) voltage, temperature, pressure, etc	
	c) voltage and current	
	d) voltage and resistance	

28.	9.In a DVM, a transducer converts *	2 points
	Mark only one oval.	
	a)input to a proportional current	
	b) input to a proportional power	
	c) input to a proportional voltage	
	d) input to a proportional resistance	
20	10 Outrout in digital forms on s't be used directly.	
29.	10.Output in digital form can't be used directly *	2 points
	Mark only one oval.	
	a) True	
	b) False	
30.	11. CRO stands for*	2 points
	Mark only one oval.	
	Cathode Ray Oscilloscope	
	Current Resistance Oscillator	
	Central Resistance Oscillator	
	Capacitance Resistance Oscilloscope	
01	40 C D O	
31.	12.C.R.O gives *	2 points
	Mark only one oval.	
	actual representation	
	visual representation	
	approximate representation	
	incorrect representation	

32.	13.Oscilloscope is *	2 points
	Mark only one oval.	
	a ohmmeter	
	an ammeter	
	a voltmeter	
	a multimeter	
33.	14.Typically oscilloscope represents*	2 points
	Mark only one oval.	
	current and time	
	resistance and time	
	voltage and time	
	power and time	
34.	15. CRO is a *	2 points
	Mark only one oval.	
	fast x-y plotter	
	slow x-y plotter	
	medium x-y plotter	
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r0000111 a001 door/200gmal.com 10000111 a100 palametrik/100gmal 100001111 as 10 habentrik/mal.com	16-27 Retectaments CRZ 26-27 Palasi Sant Minis CRZ 16-27 Retectament States CRZ	015000000 None of the above Presence of , OHO group Ninder TTROBUSES 2.4 deleterable julgated Presence of , COOM gas blade 2000-00000 None of the above Presence of , OHO was blade	Non-seally Plants planetes of price Albregon Kareanne Non-seally Plane I may be exhibited by and \$10 of the plane State Plane I may be exhausted and \$10 of the plane	arrow prosp Presence of armelia and Presentine inplication is present. Value ability in prosp. Both Armenical share shales. Greaterable in yearest. Value ability in cross. Both Presentine indicates the present. Value	emmonal shor shale ng2 hijahdadan quellatin masurment Paddah knoon. Bisa ayanna Culful ng2 at hijahdadan quellatin masurment Paddah knoon. Bisa Na et Karanda	John mellosi John mellosi John mellosi	11 800	Co-philip (Sta ACC) Restoring agent of Co-philip (Sta ACC) Restoring agent of Co-philip (Sta ACC) Report from days	Manageriale Manager Manager	. 040 . 000 . 000H	OKLODO Moral financia OKLODO Moral paid (Ten. paid 800 OKLODO Moral financia
T0000111 sa 16 hahatharge (@grad ss T00001111 sa 30 valikushaka000@gra	27/27 Toke Remark Harger PCM 91/27 Valdes Rentscaline English	GSSCF COSIS 2 A distinguismy language for recovery of COOM game, Mindele GST2 HSDUS 2 A distinguismy language for Presence of COOM game, Mindele GST2 HSDUS 2 A distinguismy languism Presence of COOM game, Mindele	Non-sonly Flame I may be under pix and \$10 of the above Kareanne Non-sonly Flame places of satisfact of the above Kareanne Non-sonly Flame	distriya prop Prasmos el armatis arei Armatis alam siste i Desalunion is prasmi. Yeline distriya prop Nore el libra distre. P Anancino l'apinolistica Companii a sermalio. Yeline	agrama Calife spiritularies quantitate manarement Related from the Nation State Spiritularies quantitate manarement Valence Valence	John method Vacametric method	1 1 N/O	[Co (MAC) ( Std ) ADD	Raterials Selection	040 000H	GKLCDR00 Annual paid / Then, year 8:00 GKLCDR00 None of the above
100001 it abit amerepaya@grad.com 100001 it abit aetahtean@grad.com	16/27 Ponel Intellecence CBZ 16/27 Adul Djob Nouve CBZ	6000 6077 Nove of the above Presence of COOR year blocks 6000 6000 NoVCOO Presence of COOR years blocks and the coordinate of the co	Non-westy Flores II may be amine All of the above Kareanne Body Flores II may be amine Allmagen Albame	althyle group Nove of the alone No. 1 Electronic Construction is present. Visited althyle group. Presence of annuals and No. 1 Electronic Construction is present. Pleas	Na or Kierlende up I of Psylvalanian querillativa manurment Particle innover Blue appears Cullida up I of Internation querillativa manurment Blue Blue	John method Solubility method	2 1 N(4) 1 N(4)	Co photog Std AGD Reducing agent of Co photog Std AGD Containing agent	Totalestate -1 Mendertate	- 000H - 000H	OKLORED Name of the place OKLORED Annual paid / Theo. year 6/00 OKLORED Name of the Paid / Theo.
roses) (1.605) katekapatikologipusis 10000) (1.663) valitarasine (1800)pus 10000) (1.663) pasantamis(2000)pus	20/27 National sharker batteryCRZ 20/27 National Porter PCRI (7/27 Projektalanian GeograficCRZ	additional forms of the above Pressure of JCDOH group bloads 06016 billion 2 plainting-beng hydroxin Pressures of JCDOH group bloads 08016 billion Configuration Pressures of JCDOH group bloads	Service Employ and the place of the Service Konsense Sonly Flores Employ and the soid this of the above Konsense Sonly Flores Employ as soleton do not the little of the above	ammy a group Bath Posserial from the State of St	na or namene splitfeldeden Beit Beitigkeit vom Bise agenne Cellick splitfeldeden sperifielde manurment Beitigkeit vom Bise ammelsel überskeite splitfeld printerlaten manurlatien manurment Beitigkeit vom	John medical John medical John medical	14 819	pur perili ( 604 PGC ) Nove of the above 10 (Car ( NOV) ( 604 PGC ) Nove of the above 10 (Car( NOV) ( 604 PGC ) Tradesires assets 10	Ratespa Ratespa Ratespa	. GHG . GHG	OKLODED Anial plot / Then, yell (100 OKLODED Ani
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nd0301120106 vasimanishasid016gg 100301120108 vasimanishid16ggmin 100301120108 vasimanishid16gmin	16 127 Valebrasi punsibalian NSPCH 26 127 Bartya panin adi shakin PCH 21 127 Valebrasi Burasi Daniala "***	ESSECTION 2 Administrational Institute Presence of GOOH group bloads OSICISSION 2 Administration Presents of arthresping bloads -0457500000 National Presence of GOVERNMENT bloads	Non-monthy Plantes — It may be continued and \$100 of the alarma. Konsenter Non-monthy Plantes — It may be arrived — 200 of the alarma. Konsenter Non-month Plantes — It may be arrived as and \$100 of the alarma.	amonym group Bash American American share shade Companed a american ability to group Procures of american information share shade Grandwish in years of Vales ability to group Bash American share shade Grandwish in years of Vales	asymme Cultid splittphridualism Basin White Street asymmetric Cultid splittphridualism specialism measurement Particle Income Black asymmetric Cultid splittphridualism specialism representation (Residualism Street, Cultid	Pyrometric realized John method John method	1.0 (a)(a) 1.1 (a)(a)	(Currently Std 160 Hydrolog agent (Currently Std 160 Nove of the above of Currently Std 160 Nove of the above	J Totalerine Edition	. 040 . 040	ORLGBOO Annal paid (Then, paid 600 ORLGBOO Annal paid (Then, paid 600 ORLGBOO Annal paid (Then, paid 600 ORLGBOO Annal paid 6 Then paid 600 ORLGBOO
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100001124541 valuture0018@gradio 100001130147 praksitele@gradiose	20/27 Value Ray Ramon CRZ 26/27 Profestioned Ceste CRZ	GREATURE 2 Administracy Inglished Presence of COOM years Minde CREATURE 2 Administracy Inglished Presence of COOM years Minde	Non-sonly Flores almost an extraction and the son of the son Karasana Non-sonly Flores I may be carbon for and \$100 of the stones Karasana	dishiple group Bath American share have Unaccurate a present Value dishiple group Presente of armatic and American share shale. Unaccurate a present Value		John method John method	1.1 N/S	Cop (ACC)   Colo   ACC   Coloniaring agent   Cop (ACC)   Coloniaring agent   Cop (ACC)   Coloniaring agent   Cop (ACC)   Coloniaring agent   Cop (ACC)   Coloniaring   Cop (ACC)   Cop (ACC)   Coloniaring   Cop (ACC)	Manageriale Relevate	010	GKLING Annal paid / Then, year Kinds GKLING Annal paid / Then, year Kinds
10030113-000 municipant00@gmaile 10030113-001 pratingentnis00@gma	23/27 Wood Span PCW 18/27 Publica jaminus CRZ	TORROLDER 2 J. Selectophory) hydracin Presence of GDGH group Ninete TORROTOR 2 J. Selectophory) hydracin Presence of unbelongston Ninete 1984 (1997) 3 J. Selectophory) hydracin Presence of unbelongston Ninete	Non-sonty Planne II may be carbonyle and Philogens Konsone Early Planne II may be carbonyle and \$40 of the above Konsone			John method Pyrometric method	1.1 (a)(a) 1.1 (a)(a)	Curpose   (00 / 00)   Reducing agent   0  Curpose   (00 / 00)   Rydesing agent	Statement	.040	OKLORGE Anal paid /Then, yell (100 OKLORGE) Anal paid /Then, yell (100 OKLORGE)
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10130113133 valnerjehi0@prel r	91/27 Vashneri partitinal CRZ	AGCETORIC Carlylanine Presence of CRO group Ninde	Sonly Flore phonon of subscyle and Richards Names	alidných group – Prosence of alighadu amic Prosentine hydrochicológica construction is present federa	No or Kardender - Ignifelder - Ignifelder manuscher Michiganischen - Michiganische - Michiganischen - Michiganische - Michiganischen - Michiganische	John method	11 (4)	(Co (MOC)) (SSE POD Containing agent 1	Menderale	.040	OSSNG Transpart School year COS

Name of Studente	Obtained Marks	Total Mark	Doroontogo
Name of Students	Obtained Marks		
Akash Vikas Dadmal Akash Vikas Dadmal	22	30	73.33333333
Akash Vikas Dadmal	19	32	59.375
	28	40	70
Akash Vikas Dadmal	45	60	75
Ankit Rajendra Telase	19	30	63.33333333
Ankit Rajendra Telase	13	32	40.625
Ankit Rajendra Telase	47	60	78.33333333
Ashwini Ashok kalaskar	22	30	73.33333333
Ashwini Ashok kalaskar	16	30	53.33333333
Ashwini Ashok kalaskar	24	32	75
Ashwini Ashok kalaskar	26	40	65
Ashwini Ashok kalaskar	49	60	81.66666667
Ashwini Banduji Tajne	20	30	66.6666667
Ashwini Banduji Tajne	27	30	90
Ashwini Banduji Tajne	30	30	100
Ashwini Banduji Tajne	13	32	40.625
Ashwini Banduji Tajne	30	32	93.75
Ashwini Banduji Tajne	24	40	60
Ashwini Banduji Tajne	37	40	92.5
Ashwini Banduji Tajne	42	60	70
Ashwini Banduji Tajne	55	60	91.66666667
Bhavna vijay chamate	19	32	59.375
Bhavna vijay chamate	30	40	75
Bhavna vijay chamate	36	60	60
Briavria Vijay Chamate	30	00	
Kalyani Dattatray Shingr	23	30	76.6666667
Kalyani Dattatray Shingr		32	87.5
kalyani Dattatray Shingr	32	40	80
Kalyani Dattatray Shingr	52	60	86.6666667
	<u> </u>	30	22.232220
Kalyani Rajendra Labhar	21	30	70
Kalyani Rajendra Labhar		32	71.875
Kalyani Rajendra Labhar	24	40	60
Kalyani Rajendra Labhar	45	60	75
Ku. Pratiksha Ashok Nar	24	30	80
Nubali kiahar Dumakan	40	20	F^
Nuhali kishor Durugkar	16	32	50
Nuhali kishor Durugkar	18	40	45
Nuhali kishor Durugkar	17	60	28.33333333
Prajkta vinayak narad	17	30	56.6666667
Prajkta vinayak narad	28	30	93.33333333
prajkta vinayak narad	21	30	65.625
prajnia villayan ilalau		32	00.025

Prajkta vinayak narad	19	40	47.5
Prajkta vinayak narad	43	60	71.66666667
Prajkta villayak ilarau	43	60	11.00000001
Pratiksha Ashok Narawa	25	32	78.125
Pratiksha Ashok Narawa	33	40	82.5
Pratiksha Ashok Narawa	45	60	75
Platiksila Asilok Nalawa	45	60	75
Pratiksha Nandkishor Ka	25	30	83.33333333
Pratiksha Nandkishor Ka	24	30	80
Pratiksha Nandkishor Ka	23	32	71.875
Pratiksha Nandkishor Ka	26	32	81.25
Pratiksha Nandkishor Ka	27	40	67.5
Pratiksha Nandkishor Ka	36	40	90
Pratiksha Nandkishor Ka	35	40	87.5
Pratiksha Nandkishor Ka	45	60	75
Pratiksha Nandkishor Ka	43	60	70
Platiksila Naliukisiloi Ka	42	60	70
Preeti deepak datta	36	60	60
Preeti deepak dutta	29	30	96.6666667
Preeti deepak dutta	22	30	73.33333333
Preeti Deepak Dutta	17	32	53.125
Preeti deepak dutta	21	32	65.625
Preeti deepak dutta	30	32	93.75
Preeti deepak dutta	16	32	50
Preeti deepak dutta	23	40	57.5
Preeti deepak dutta	40	40	100
Preeti deepak dutta	28	40	70
Preeti deepak dutta	60	60	100
Preeti deepak dutta	45	60	75
•			
Priya Ramesh Dhande	23	30	76.6666667
Priya Ramesh Dhande	15	30	50
Priya Ramesh Dhande	21	32	65.625
Priya Ramesh Dhande	26	32	81.25
Priya Ramesh Dhande	26	40	65
Priya Ramesh Dhande	40	40	100
Priya Ramesh Dhande	44	60	73.33333333
puja suresh Telmasre	13	30	43.33333333
Puja suresh Telmasre	19	32	59.375
puja suresh Telmasre	20	32	62.5
Puja suresh Telmasre	27	40	67.5
puja suresh Telmasre	27	40	67.5
Puja suresh Telmasre	44	60	73.33333333
puja suresh Telmasre	28	60	46.6666667
Rakhi Prakash Sakharka	22	30	73.33333333
Rakhi Prakash sakharkai	29	30	96.6666667
Rakhi Prakash Sakharka	24	32	75
Rakhi Prakash Sakharka	33	40	82.5
Rakhi Prakash Sakharka	45	60	75

Rani Bandi Sahare	60	60	100
Rani Bandu Sahare	19	30	63.33333333
Rani Bandu Sahare	30	30	100
Rani Bandu Sahare	23	32	71.875
Rani Bandu Sahare	32	32	100
Rani Bandu Sahare	29	32	90.625
Rani Bandu Sahare	21	32	65.625
Rani Bandu Sahare	32	32	100
Rani Bandu Sahare	22	40	55
Rani Bandu Sahare	40	40	100
Rani Bandu Sahare	38	60	63.33333333
Sagar Gopichand Behad	22	30	73.33333333
Sagar Gopichand Behad	21	30	70
Sagar Gopichand Behad		32	62.5
Sagar Gopichand Behad		32	71.875
Sagar Gopichand Behad	29	40	72.5
Sagar Gopichand Behad	26	40	65
Sagar Gopichand Behad	43	60	71.66666667
Sagar Gopichand Behad	31	60	51.66666667
Cagar Copionaria Bonda	0.		01100000001
Sakshi raju bakane	40	40	100
Sakshi Raju Bakane	26	30	86.6666667
Sakshi raju bakane	28	30	93.33333333
Sakshi raju bakane	30	30	100
Sakshi raju bakane	30	30	100
Sakshi raju bakane	23	23	100
Sakshi raju bakane	31	31	100
Sakshi raju bakane	32	32	100
Sakshi raju bakane	32	32	100
Sakshi raju bakane	32	32	100
Sakshi raju bakane	29	40	72.5
Sakshi raju bakane	40	40	100
Sakshi raju bakane	40	40	100
Sakshi raju bakane	51	60	85
Sakshi raju bakane	59	60	98.33333333
Sakshi raju bakane	58	60	96.66666667
Sakshi raju bakane	59	60	98.33333333
Oaksiii raja bakane	33	- 00	30.3333333
Sarika Madhukar Jiotode	25	30	83.33333333
Sarika Madhukar Jiotode		30	93.33333333
Sarika Madhukar Jiotode		30	100
Sarika Madhukar Jiotode		32	65.625
		32	93.75
Sarika Madhukar Jiotode Sarika Madhukar Jiotode		32	93.75
Sarika Madhukar Jiotode		32	100
Sarika Madhukar Jiotode		40	82.5
Sarika Madhukar Jiotode		40	97.5 97.5
Sarika Madhukar Jiotode		40	97.5
Sarika Madhukar Jiotode	46	60	76.66666667
Sarika Madhukar Jiotode	50	60	83.33333333

Shraddha dhote	17	30	56.6666667
Shraddha dhote	27	32	84.375
Shraddha dhote	32	32	100
Shraddha dhote	31	32	96.875
Shraddha dhote	34	40	85
Shraddha dhote	42	60	70
Shraddhadhote13@gmai	32	32	100
om addition to roughia	<u> </u>	<u> </u>	100
Shruti Prakash Neware	16	30	53.33333333
Shruti Prakash Neware	25	30	83.33333333
Shruti Prakash Neware	30	30	100
Shruti Prakash Neware	14	32	43.75
Shruti Prakash Neware	30	32	93.75
Shruti Prakash Neware	31	32	96.875
Shruti Prakash Neware	32	32	100
Shruti Prakash Neware	15	40	37.5
Shruti Prakash Neware	35	40	87.5
Shruti Prakash Neware	40	40	100
Shruti Prakash Neware	22	60	36.66666667
Shruti Prakash Neware	41	60	68.33333333
Shruti Prakash Neware	54	60	90
Omuti i iakasii Neware	37	00	30
Uma Madhav Bhanarkar	21	30	70
Uma Madhav Bhanarkar	30	30	100
	30	30	
Uma Madhay Bhanarkar			100
Uma madhay Bhanarkar	24	32	75
Uma madhav Bhanarkar	32	32	100
Uma Madhav Bhanarkar	31	32	96.875
Uma Madhav Bhanarkar	26	40	65
Uma Madhav Bhanarkar	40	40	100
Uma Madhav Bhanarkar	39	40	97.5
Uma Madhav Bhanarkar	39	60	65
Uma Madhav Bhanarkar	60	60	100
Uma Madhav Bhanarkar	60	60	100
Uma Madhav Bhanarkar	60	60	100
Varsha Pandurang Gowa		30	73.33333333
Varsha Pandurang Gowa		32	84.375
Varsha Pandurang Gowa		40	70
Varsha Pandurang Gowa	53	60	88.33333333
	_		
Vijeta Vasantarao Darol	51	60	85
Vijeta Vasantarao Daroka	22	30	73.33333333
Vijeta Vasantarao Daroka	30	30	100
Vijeta Vasantarao Daroka	22	32	68.75
Vijeta Vasantarao Daroka	32	32	100
Vijeta Vasantarao Daroka	28	40	70
Vijeta Vasantarao Daroka	39	40	97.5
Vijeta Vasantarao Daroka	60	60	100

# Gondwana University Gadchiroli B.Sc. Zoology Semester V Summer 2021 Online Practical Examination

122 responses

	ana	

Name of Student (Surname, Name, Middle Name)

122 responses

Divyani Sanjiv Wasekar

Megha Bandu Nannaware

Rohit vinod pongde

Nikhita Sureshrao Dharmik

Prachi. B. Bhujade

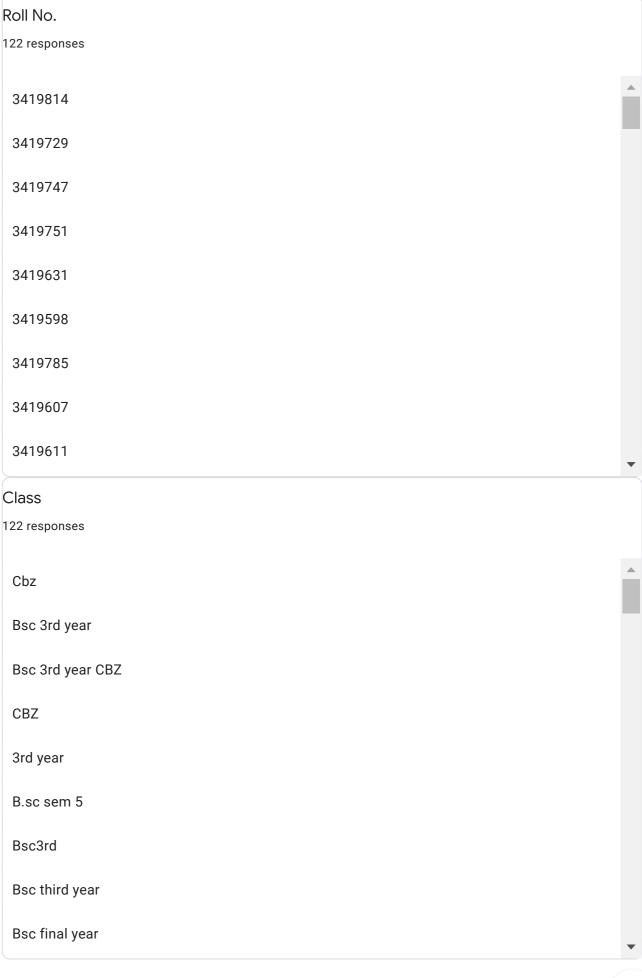
Nishigandha Tonge

Revati babakar burande

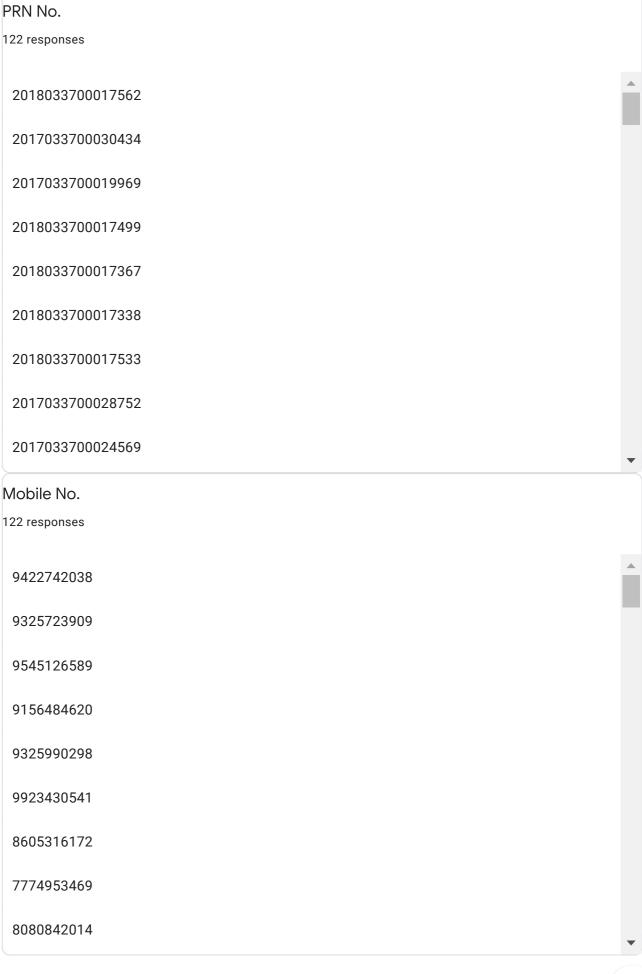
Sanjiwani Suresh Betal

Anjali sanjay guntiwar

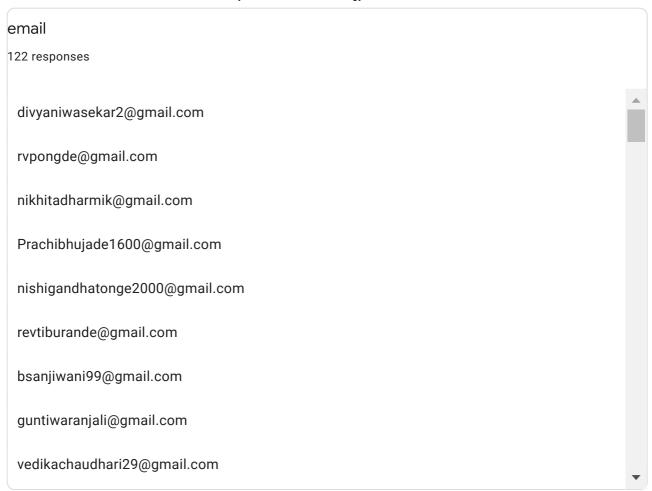




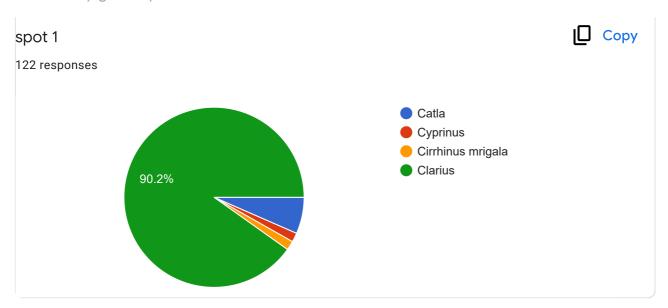




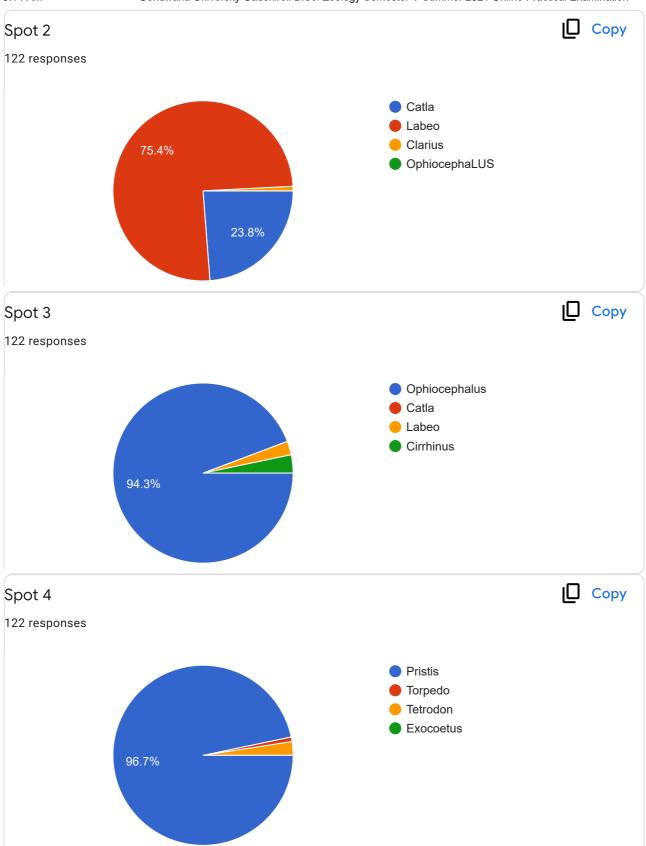




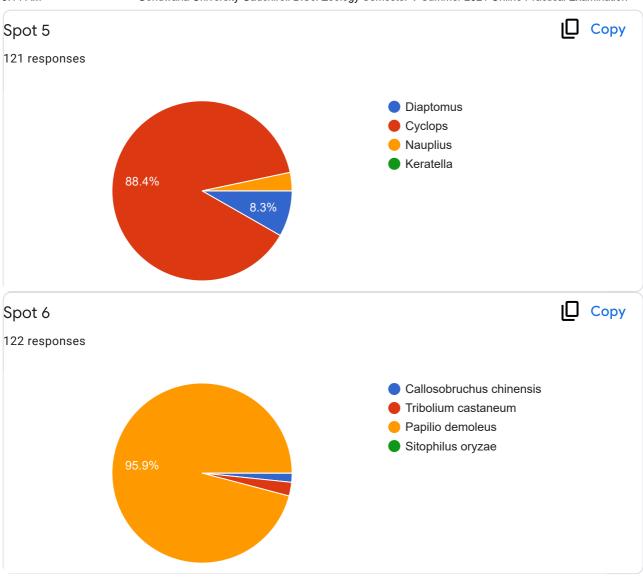
#### Identify given spot



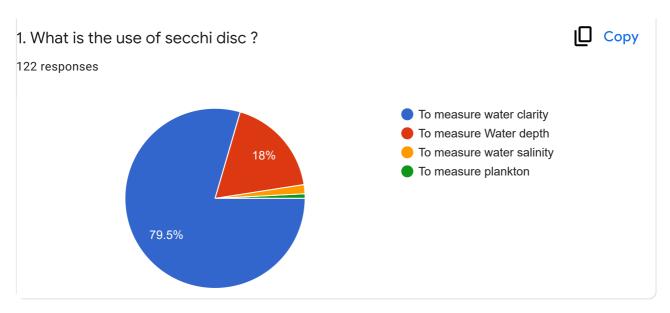




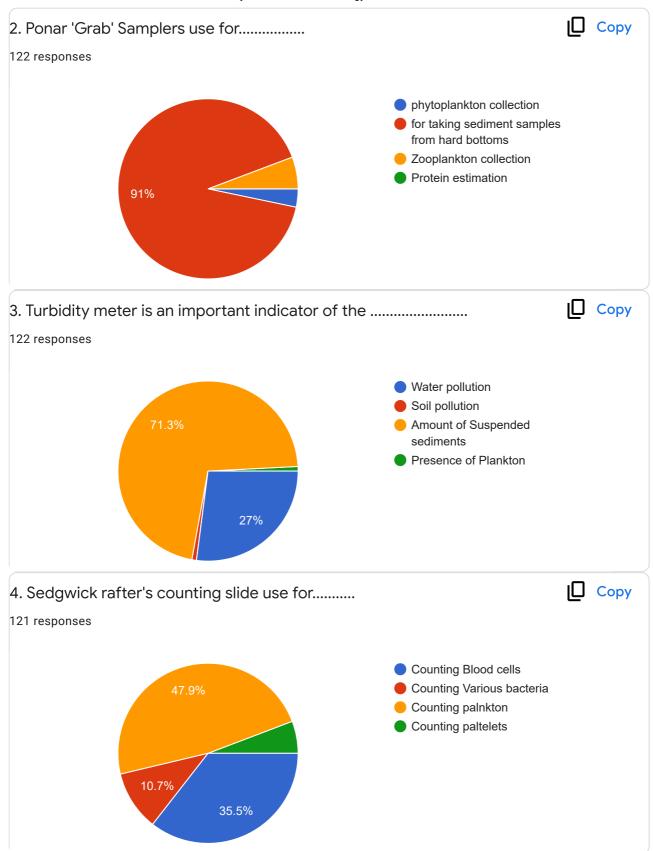




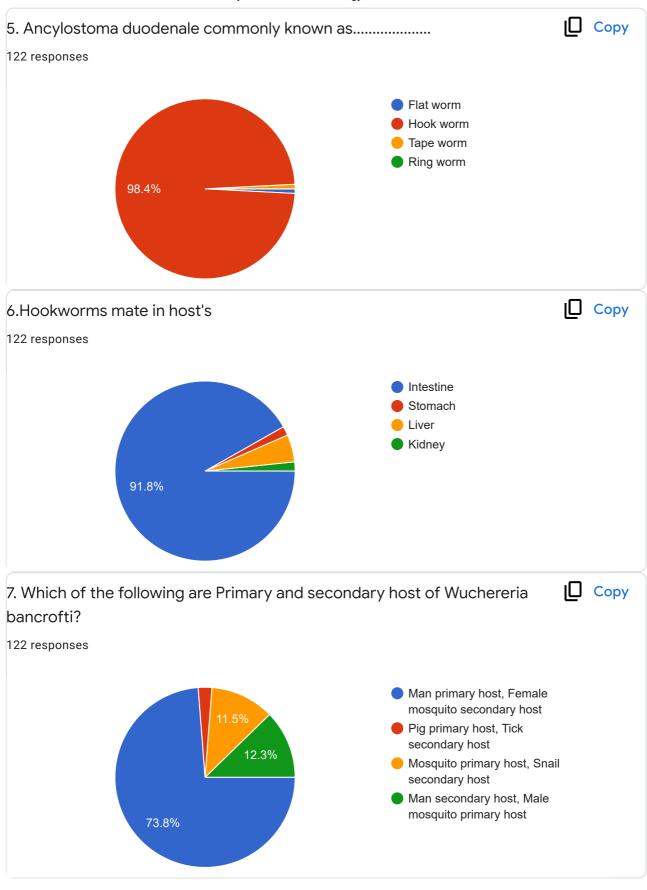
#### **MAJOR EXPERIMENTS**





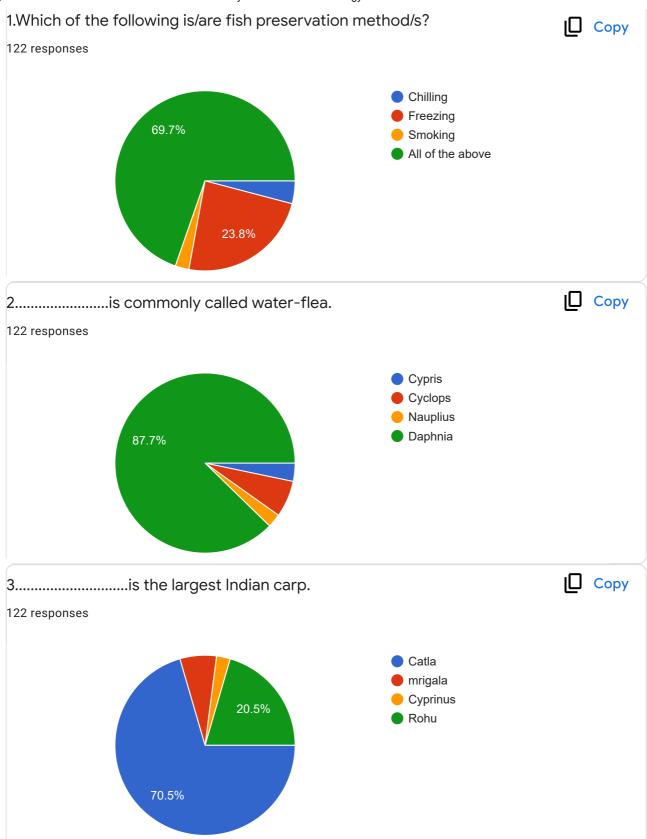




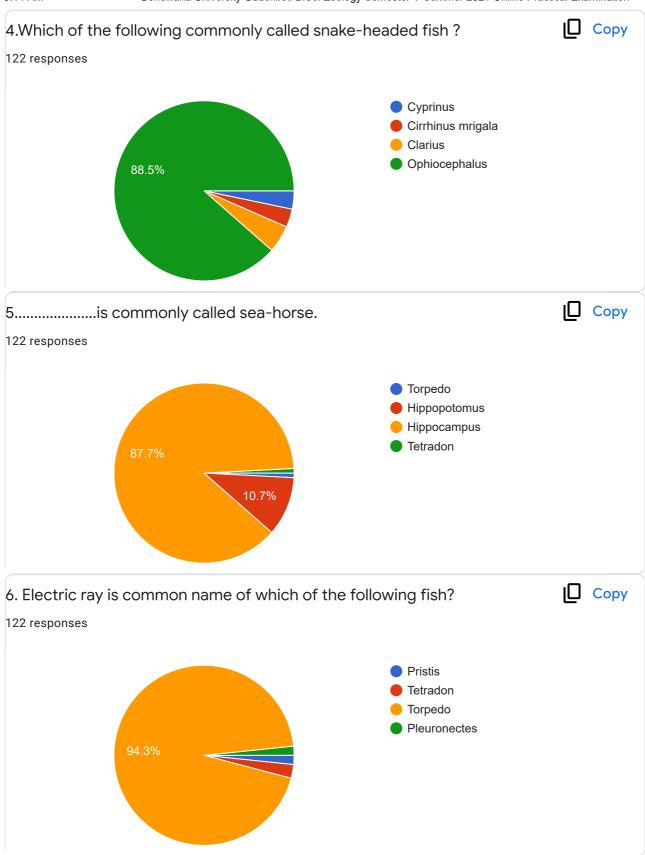


Viva









**Practical record Submission** 

Have you submitted practical record?
83 responses

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# Google Forms



# Gondwana University Gadchiroli B.Sc. Zoology Semester V Summer 2021 Online Practical Examination

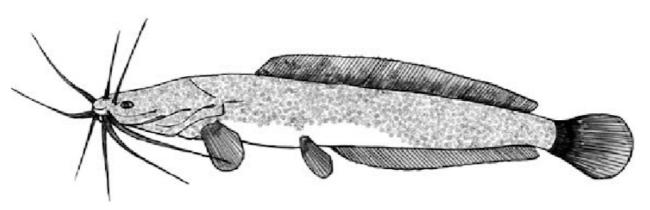
Gondwana University Gadchiroli

	Practical Exam Question-Answer Format [As per the Gondwana University Practical Examination Scheme]	
	Centre: Anand Niketan College, Anandwan, Warora.	
	Time: 50 Minutes	
	Marks: 30	
*	Required	
1.	Email *	
2.	Name of Student (Surname, Name, Middle Name) *	
3.	Roll No. *	
4.	Class *	
5.	PRN No. *	

- 6. Mobile No. \*
- 7. email \*

Identify given spot

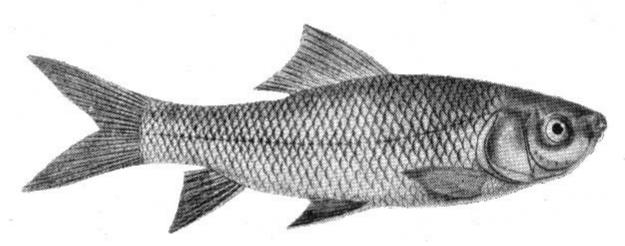
8. spot 1\*



/	- 1	C -+	_
(	)	Cat	ıa

- Cyprinus
- Cirrhinus mrigala
- Clarius (

9. Spot 2 \*

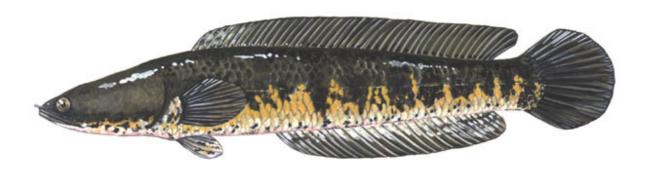


Mark only one oval.

1	0-4	۱.
	Cat	la

- Labeo
- Clarius
- OphiocephaLUS

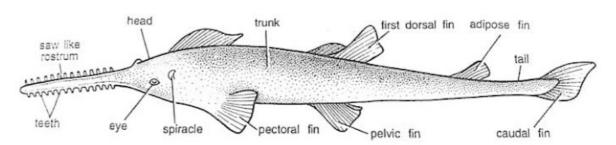
10. Spot 3 \*



/	\ ~		•			
1	1 (1	nh.	-	eph	ากเ	110
\	<i>/</i> <b>U</b>	ווט	IUC	CUI	ıaı	นอ

- Catla
- Labeo
- Cirrhinus

## 11. Spot 4 \*

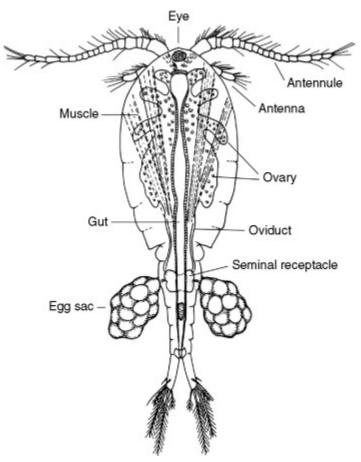


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(	- /	Ρr	เธเ	15

		_	
(	)	lorpe	d٥
/		TOIPE	uu

- Tetrodon
- Exocoetus

## 12. Spot 5



- Diaptomus
- Cyclops
- Nauplius
- ( Keratella

## 13. Spot 6 \*



Mark only one oval.

( ) Callosobruchus chinensis	າຣiຣ
------------------------------	------

, ilibolialli castalicali		Tribolium	castaneum
---------------------------	--	-----------	-----------

(	) Papilio	demo	وربطا
(	) i apino	ucillo	icus

	Sitophilus	oryzae
\		- · <b>,</b> — · · ·

#### **MAJOR EXPERIMENTS**

14.	1	\//ha+	ic tha	uco of	secchi	dica	2 *
14.	١.	vvnai	us une	use or	secon	CHSC	• •

Mark only one oval.

	_			
)	IΩ	measure	water	clarity
. /		IIICUSUIC	WULCI	Cidiity

To measure Water depth

To measure water salinity

To measure plankton

15.	2. Ponar 'Grab' Samplers use for
	Mark only one oval.
	phytoplankton collection for taking sediment samples from hard bottoms Zooplankton collection Protein estimation
16.	3. Turbidity meter is an important indicator of the*
	Mark only one oval.
	Water pollution Soil pollution Amount of Suspended sediments Presence of Plankton
17.	4. Sedgwick rafter's counting slide use for
	Mark only one oval.
	Counting Blood cells
	Counting Various bacteria
	Counting palnkton  Counting paltelets
18.	5. Ancylostoma duodenale commonly known as*
	Mark only one oval.
	Flat worm
	Hook worm
	Tape worm
	Ring worm

19.

6. Hookworms mate in host's

	Mark only one oval.
	Intestine
	Stomach
	Liver
	Kidney
20.	7. Which of the following are Primary and secondary host of Wuchereria bancrofti? *
	Mark only one oval.
	Man primary host, Female mosquito secondary host
	Pig primary host, Tick secondary host
	Mosquito primary host, Snail secondary host
	Man secondary host, Male mosquito primary host
Viv	/a
21.	1.Which of the following is/are fish preservation method/s? *
	Mark only one oval.
	Chilling
	Freezing
	Smoking
	All of the above

22.	2s commonly called water-flea. *
	Mark only one oval.
	Cypris
	Cyclops
	Nauplius
	Daphnia
23.	3is the largest Indian carp. *
	Mark only one oval.
	Catla
	mrigala
	Cyprinus
	Rohu
24.	4.Which of the following commonly called snake-headed fish?*
	Mark only one oval.
	Cyprinus
	Cirrhinus mrigala
	Clarius
	Ophiocephalus
25.	5is commonly called sea-horse. *
	Mark only one oval.
	Torpedo
	Hippopotomus
	Hippocampus
	Tetradon

26.	6. Electric ray is common name of which of the following fish? *
	Mark only one oval.
	Pristis
	Tetradon
	Torpedo
	Pleuronectes
Pra	actical record Submission
27.	Have you submitted practical record ?
	Files submitted:

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