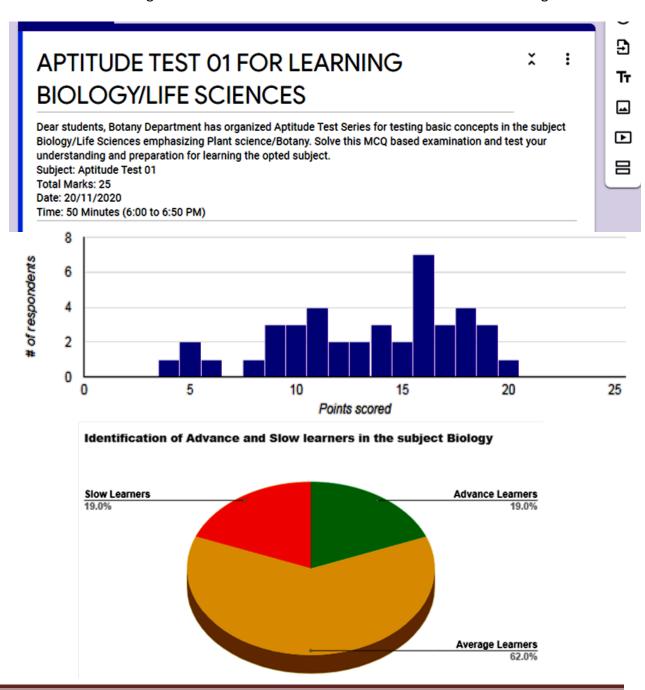
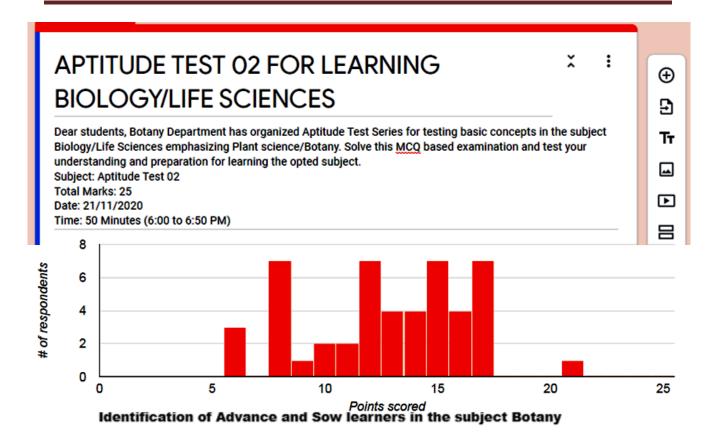
Department of Botany

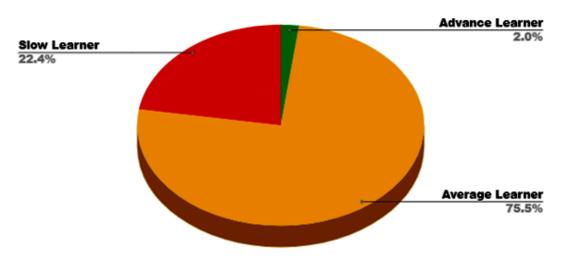
Identification of advanced and slow learners in the subject Botany Session 2020-21

Criteria used for identification of Advanced (Bright) and Slow (Weak) learners:

- General Aptitude Tests were (two consecutive tests) conducted at the beginning of the session.
- Students scoring above 70% marks in both the tests were identified as bright students.
- Students scoring below 40% marks in both the tests were identified as weak students.
- Students scoring between 41% to below 69% marks were identified as average students.







Links for the Aptitude Tests conducted for the identification of advanced and slow learners during the academic session 2020-21:

Aptitude Test 01

https://docs.google.com/forms/d/1hVe9fW0WtuqX0SFtxL16pWCCgn8lhLGSsmJedsxAuRM/edit?usp=sharing

Aptitude Test 02

https://docs.google.com/forms/d/1fB0GT1gcAlEIUJbB3swRudpx RKJkz5M5MmziXmMGM/edit?usp=sharing

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Activities conducted for advanced learners:

- Encouraged students for model, chart and rangoli making based on the syllabus topics learnt in the classroom.
- Encouraged students for participation in Symposium, Seminar, Conference and various completions like quiz, poster and photography organized by the college and other institutions.
- Provided guidance for identification of their interests and career planning through personal communication.



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Efforts taken for Average and Slow Learners

- Run Bridge Courses prior to teaching topics in the prescribed syllabus.
- In person follow up for class attendance, assignment completion and problem solving.

Links for reading materials provided in Bridge Courses

- https://drive.google.com/file/d/1yt9QUNt1HGY pnrgkVXkqIzF3KNDsQ9/view?usp=sharing
- https://drive.google.com/file/d/1AcLajyRvtwFDifX5oHXqQvqYadK6LqG/view?usp=sharing
- https://drive.google.com/file/d/1QXlw7YSw60Dwbmc0JKWEwRHSAzJE3tXQ/view?usp=sharing



Prashant Wagh

Jun 20, 2020

Dear students,

Sharing here is the reading material under bridge course for linking basics and fundamentals of the topics mentioned in the semester III syllabus.

Go through the reading material and post your doubts and queries.



Bridge course material PI...

PDF



Prashant Wagh

Jun 18, 2020 (Edited Jun 19, 2020)

Dear students,

Posting here is the reading material for recalling the basics from 12th standard. Go through it and post any query/doubts raise during reading.



Molecular Basis of inherita...

PDF



Prashant Wagh

Apr 5, 2021

Dear students,

uploaded here is the portion from 12th Standard Biology Textbook where some basic aspects of the Plant Biotechnology {11.3 Tissue Culture (page number 249-252) and 12.4 Application of Biotechnology D. Genetically Modified Organisms (page number 282 - 285)} are mentioned. Hope you all have studied this portion in undergraduate classes. You are suggested to go through these topics once again in order to improve your understanding further at detailed level during the current semester. Queries/repsonses are eagerly awaited in the comment box.



Bridge Course SEm VI Rea...

PDF

Department of Chemistry CHEMISTRY TEST EXAM FOR SLOW

LEARNER STUDENTS

Department of Chemistry, Anand Niketan College Warora

	<u> </u>	
*	Required	
1.	Email *	
2.	FULL NAME *	
3.	CLASS *	
	Mark only one oval.	
	PCM [B.Sc 1 st year]	
	CBZ [B.Sc 1 st year]	
Q	UESTIONS	ALL QUESTIONS ARE COMPULSORY. EACH QUESTION CARRY 2 MARKS.
4.	1] The element, with atomic number 118, will be *	
	Mark only one oval.	
	alkali	
	onoble gas	
	lanthanide	
	transition element	

5.	2] Cation and anion combines in a crystal to from following type of compound.
	Mark only one oval.
	ONIC
	METALLIC
	COVALENT
	DIPOLE-DIPOLE
6.	3] The angular shape of ozone molecule (O3) consists of : *
	Mark only one oval.
	1 sigma and 2 pi bonds
	2 sigma and 2 pi bonds
	1 sigma and 1 pi bonds
	2 sigma and 1 pi bonds
7.	4] Which of the following species has four lone pairs of electrons? *
	Mark only one oval.
	<u> </u>
	CI-
	He

8.	5] Which structure is linear? *
	Mark only one oval.
	\$02
	CO2
	CO3
	S04
9.	6] Which statement is NOT correct? *
	Mark only one oval.
	$\hfill \Box$ A sigma bond is weaker than a π-bond.
	$\hfill \Box$ A sigma bond is stronger than a π-bond.
	A double bond is stronger than a single bond
	A double bond is shorter than a single bond
10.	7] Strongest hydrogen bond is shown by *
	Mark only one oval.
	water
	Ammonia
	Hydrogen fluoride
	Hydrogen sulphide.

11.	8] The correct order regarding the electronegativity of hybrid orbitals of carbon is *								
	Mark only one oval.								
	sp > sp2 > sp3								
	sp < sp2 > sp3								
	sp < sp2 < sp3								
	sp > sp2 < sp3								
12.	9] Lassaigne's test in qualitative analysis to detect *								
	Mark only one oval.								
	Nitrogen								
	Sulphur								
	Chlorine								
	All of these								
13.	10] Among the elements Ca,Mg,P and Cl, the order increasing atomic radii is : * Mark only one oval.								
	Ca <mg<p<cl< td=""></mg<p<cl<>								
	Mg <ca<cl<p< td=""></ca<cl<p<>								
	CI <p<mg<ca< td=""></p<mg<ca<>								
	P <ci<ca<mg< td=""></ci<ca<mg<>								

14.	11] Which one of the following ions will be the smallest in size? *
	Mark only one oval.
	Na+ Mg(2+) F- O(-2)
15.	12] Among the following oxides, the one which is most basic is *
	Mark only one oval.
	ZnOMgOAl203N205
16.	13] Elements of which of the following groups will form anions most readily? * Mark only one oval. Oxygen family Nitrogen family Halogens Alkali metals

17.	14] In the periodic table, with the increase in atomic number, the metallic character of an element *
	Mark only one oval.
	Decreases in a period and increases in a group Increases in a period and decreases in a group Increases both in a period and the group Decreases in a period and the group.
18.	15] The oxidation number of chromium in potassium dichromate is *
	Mark only one oval.
	<u>+6</u>
	5
	2
	+2
19.	16] The low density of ice compared to water is due to *
	Mark only one oval.
	hydrogen-bonding interactions
	dipole-dipole interactions
	dipole-induced dipole interactions
	induced dipole-induced dipole interactions

20.	17] Be(2+) is isoelectronic with which of the following ions? *
	Mark only one oval.
	H+
	Li+
	Na+
	Mg(2+)
21.	18]are saturated hydrocarbon, represented by general formula CnH2n+2. *
	Mark only one oval.
	Alkanes
	Alkyl halides
	Alkyl groups
	d. Alcohols
22.	19] Which one of the following reagent react with alcohol to give alkyl halides *
	Mark only one oval.
	Halogen acids
	Phosphorus halide
	Thionyl chloride
	All of the above

26.	23] When alkyl halide is treated with sodium alkoxide the product is *
	Mark only one oval.
	alcohol
	ether
	ester
	ketone
27.	24] Picric acid is- *
	Mark only one oval.
	2,4-dinitrophenol
	2,6-dinitrophenol
	2,4,6-trinitrophenol
	2,3,5-trinitrophenol
28.	25] Williamson synthesis is used to synthesize- *
	Mark only one oval.
	ester
	ether
	ketone
	aldehyde

Timedamp Email-Address Score 6/62/27 16 16 16 washin/18broad.com	FLEATHER CLASS (The element, w	th atom2) Cation and artison	conti 3). The angular shape of si 2 stores and 1 of bonds.	(Which of the following of (Which structure)	inne II (Wish sistement is NOT) Strongest hydrogen bod A size a bond is weater 10 tolonom Suntie	[] The current order rep or 7 and 7 and	er?(Lessigne's test	n qual 12 Among the elem Caroller PHC	mis (11)What one of MisQr)	the falls 12] denoing 8	he following or CE Elements of white Platestern	other \$140 in the persons labels, a 150 The social December, in a serial and 15	alian number (6) The loss density of loss (7) Be(2+) is in hadronen bondon internal (2+	electron (II) — are solunie (II) (II)	Sich one of the bills(20) (DNI reaction is	r marrier 22(Aby halides can be po21)When alby in All of the above	tide is to 20) Plant acid in 20) Williamson synthes 24.6 trickleshami when	als is used in spellenian
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692021 11 68 20 denhanatie10@gmail.co	34 / 10 Darshan Mareshaur ente CRZ (B. So 1 si prac) albeit	IONC	1 sigma and 2 pi bomis 1	D 000	A double bond is stronger Hydrogen Suntile	6 > 65 > 65 6 > 60 > 61	Alaftese	CHP-Np/Ca	M(2+)	240	Abel melab	Decreases in a period are +6	higher product spensy.	Abares Plan	house habite All of Breas Andrew	Al of the above — when	2,6,6 troinghesal when	
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683001 13 63-56 rephalos@gmail.com 683001 13 65-20 broadepopinia@gmail.com	26 / 60 Rabesh Suresh Grashar CRZ (6.5s 1 si prad) noble pas 24 / 60 Populate Separan binnake CRZ (6.5s 1 si prad) noble pas	COVALENT	2 sigma and 2 pi bombs 1 1 sigma and 2 pi bombs 1	5 804 802	A double band is shader Briphagen Sunder or A signal band is sharper Briphagen Sunder or	6 - 465 - 462 6 - 465 - 462	All of these All of these	MgrCanD+P C+P+MgrCa	P. Mg(2+)	MgO MgO	Ninger landy Helegers	Decreases in a period and +2 Increases in a period and +6	hydrogen bonding interaction dynin objects interactions. Un	Abeliables Plong Aberes Holey	Anna habir Unimiessia Bessamile er anik Unimiessia Bessamile	All of the above redex PCB absolut	2,6,6 trininghesi eller 2,6,6 trininghesi telore	
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ERSON 13-06-13 philitiffSR@gmail.com ERSON 13-06-10 dubantifpeni@gmail.com	28 / 60 Yesh barine CRZ (8.5s 1 si year) nakin yan 42 / 60 Tajasai marashaar dahar CRZ (8.5s 1 si year) nakin yan	IONC IONC	2 sigme and 1 pi bombs 1 1 sigme and 1 pi bombs 1	D 000	A signational is weather Deader	minery only minery only	All of these All of these	ChPNly/Ca ChPNly/Ca	Nat Wg(2+)	MgO MgO	Nitrogen bendy Halogena	Decreases in a period and +6 Decreases in a period and +6	hydrogen konding bilmanility(2+) hydrogen konding bilmanilit+ hydrogen konding bilmanilit+ hydrogen konding bilmanilit+ di dipole dipole bilmanilites, lit+	Above Hospi	Resione All of Besse Phenol Prince habite Unimalescalar Servanniale	PCS alsolated All of the above when	2,6 sinkephenal miler 2,6 bindruphenal silver	
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Department of Physics

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Timestamp	Email Address	Score	NAME	MOBILE NUMBER				
6/30/2021 7:43:16	bhushankamdi2018@gmail.com	18 / 20	Bhushan kamdi	9325359882				
6/30/2021 7:43:53	monughaveghave@gmail.com	6 / 20	Mohini sanjayrao ghavghave	9673086401				
6/30/2021 7:46:44	ambikapijdurkar@gmail.com	18 / 20	Ambika Vinayak Pijdurkar	9579439391				
6/30/2021 7:46:47	pratikshabadkhal2002@gmail.com	18 / 20	Pratiksha vilas badkhal	7498434622				
6/30/2021 7:48:27	pranaypatil1682001@gmail.com	6 / 20	Pranay patil	8605892809				
6/30/2021 7:48:38	varshashastrakar40@gmail.com	18 / 20	Varsha Vasanta Shastrakar	8605467935				
6/30/2021 7:50:15	mzade202020@gmail.com	16 / 20	Monika Ravindra Zade	8975420210				
6/30/2021 7:51:02	yogitataksande5@gmail.com	18 / 20	Yogita Janardhan Taksande	8669251959				
6/30/2021 7:51:45	goluaskar1999@gmail.com	14 / 20	Golu deorav askar	8412009960				
6/30/2021 7:54:21	khulsangeabhi@gmail.com	16 / 20	Abhilasha Shivshankar Khulsange	9067716791				
6/30/2021 7:54:58	vaishnavimendhe20@gmail.com	14 / 20	Vaishnavi Mendhe	7620190326				
6/30/2021 7:54:58	mahakulkarsakshi@gmail.com	2/20	Sakshi Raju mahakulkar	9322632120				
6/30/2021 7:55:30	dikshafulzele14@gmail.com	18 / 20	Dikaha Fulzele	7744957425				
6/30/2021 7:56:09	renukamilmile33@gmail.com	6 / 20	Renuka Sambaji Milmile	8421420184				
6/30/2021 7:57:03	Sumanramteke4@gmail.com	20 / 20	Renuka Devidas Ramteke	9604035215				
6/30/2021 7:59:22	aditiwalde24@gmail.com	18 / 20	Aditi walde	94218 82257				
6/30/2021 8:00:32	jayshriwakde2002@gmail.com	14 / 20	Jayshri Naresh Wakade	9579201113				
6/30/2021 8:00:35	suhasgawai2001@gmail.com	20 / 20	Suhas Gawai	9860795361				
6/30/2021 8:00:39	pratikshachide2020@gmail.com	18 / 20	Pratiksha Raju Chide	9552157390				
6/30/2021 8:04:55	vaishnvikhapne@gmail.com	8 / 20	Vaishnavi Mohan Khapane	8766420803				
6/30/2021 8:06:21	gauravmoharle2002@gmail.com	20 / 20	Gaurav Moharle	7666406998				
6/30/2021 8:07:40	borap5366@gmail.com	18 / 20	Pooja bora	8856862716				
6/30/2021 8:07:43	gopalghodmare14@gmail.com	18 / 20	Gopal	7448171846				
6/30/2021 8:09:50	gvaishnavi412@gmail.com	16 / 20	Vaishnavi Gawande	9309894188				
6/30/2021 8:10:24	ramdasaskar@gmail.com	18 / 20	Bhagyashri Ramdas Askar	7666920388				
6/30/2021 8:12:37	khiratkarlaxmi@gmail.com	14 / 20	Laxmi Raju Khiratkar	8805245054				
6/30/2021 8:13:09	mayurchikte2002@gmail.com	18 / 20	Mayur chikte	8767570368				
6/30/2021 8:14:25	ashwinipimpalshende05@gmail.com	16 / 20	Ashwini Maroti pimpalshende	9322126429				
6/30/2021 8:14:29	sb04032002@gmail.com	12 / 20	SHIVAM RANGNATH BAWANE	9511241373				
6/30/2021 8:14:37	monalilambat2001@gmail.com	16 / 20	Monali Ramdas Lambat	9359395232				
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6/30/2021 8:15:12	taleganesh56@gmail.com	16 / 20	Karishma Ganesh Tale	8767066802				
6/30/2021 8:16:27	nikhitamalekar64@gmail.com	20 / 20	Nikhita ravindra malekar	7498872918				
6/30/2021 8:17:07	madhuriawachat12@gmail.com	20 / 20	Madhuri Vitthal Awachat	7666104665				
6/30/2021 8:17:28	nikureprajwal@gmail.com	16 / 20	Prajwal sheshrao nikure	7083588136				
6/30/2021 8:18:30	meghasontakke4567@gmail.com	12 / 20	Megha Kisan sontakke	9604632565				
6/30/2021 8:25:43	shubhamdeotale16606@gmail.com	20 / 20	Shubham Yadav Deotale	8668984460				
6/30/2021 8:26:05	akankshajivtode5525@gmail.com	†	Akanksha Raju Jivtode	8379844690				
	sanketbhadke22@gmail.com	+	Sanket bhadke	7498445800				
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	vijayanannaware@gmail.com	16 / 20	Vijaya Rambhau Nannaware	7666394835				
	prashantmeshram0873@gmail.com	+	Prashnant Ramesh Meshram	7798288687				
6/30/2021 8:32:53	darekarshraddha482@gmail.com	12 / 20	Shraddha Gajanan Darekar	8767533485				
	roshanmahakulkar247@gmail.com	+	Roshan Mahakulkar	9834262053				
6/30/2021 8:38:35	nandinipachare037@gamil.com	20 / 20	Nandini narendra pachare	8767478834				

6/30/2021 8:42:36	hanwaterohan61@gmail.com	14 / 20	Rohan hanuman hanwate	8605317217
6/30/2021 8:43:45	jayshripakmode2020@gmail.com	20 / 20	jayshri suresh pakmode	9322284986
6/30/2021 8:46:00	pornimawararkar6@gmail.com	20 / 20	Pornima Rajkumar Wararkar	8767068835
6/30/2021 8:49:57	ravinaambaghare001@gmail.com	20 / 20	Ravina Ambaghare	8766458159
6/30/2021 8:51:11	detheashish53@gmail.com	20 / 20	Ashish shankar dethe	8605561755
6/30/2021 8:53:44	janavipethe6@gmail.com	16 / 20	Janavi pethe	8856819860
6/30/2021 8:53:57	sakshiwatkar66@gmail.com	4 / 20	Sakshi watkar	9420554545
6/30/2021 9:07:44	pranalizile99@gmail.com	16 / 20	Pranali G. Zile	7620895045
6/30/2021 9:33:55	vivekthak2002@gmail.com	14 / 20	Vivek nanaji thak	9359846344
6/30/2021 9:49:04	dabhekarsudhir@gmail.com	8 / 20	Sudhir Dabhekar	9923259204
6/30/2021 9:58:43	sureshgohane20@gmail.com	18 / 20	Chaitali suresh gohane	8999884205
6/30/2021 10:24:42	ksatpute5822@gmail.com	20 / 20	Kajal namdeo satpute	7218728584
6/30/2021 16:12:52	komalkamdi2001@gmail.com	14 / 20	Komal kamdi	8767257494
6/30/2021 18:34:23	vmahalle334@gmail.com	16 / 20	Vaishnavi Gajanan Mahalle	8010478871
6/30/2021 20:42:39	vbhoyar4620@gmail.com	10 / 20	Vaishnavi vitthal bhoyar	9156198529
7/2/2021 15:50:34	ankitawasekar11@gmail.com	8 / 20	Ankita	9359466559

Department of Physics Surprise test for the identification of

Slow Learner & Advanced Learner Session - 2020-2021

Class: B.Sc.I Year

Sr.n	Gender	Name of Students	Marks
4	Male	BAILMARE GANESH ARUN	5
7	Female	BHAGAT SHRUTIKA MURLIDHAR	7
15	Female	DAHULE MOHINI VINOD	6
16	Male	DAHULE GANESH RAJU	3
17	Female	DAKHARE NILIMA VILAS	4
20	Female	DARVE DHANASHRI VILAS	7
25	Female	DODAKE PRATIKSHA MURLIDHAR	6
28	Male	GHATE NAYAN PANDURANG	7
29	Female	GHAVGHAVE MOHINI SANJAYRAO	6
31	Male	GHODMARE GOPAL LOKNATH	2
32	Female	JAMBHULKAR SAKSHI MAROTI	7
38	Male	KASHTI UMESH SURESH	1
39	Male	KHADE SAGAR BANDU	5
41	Female	KHARKAR VRUSHALI VIJAY	6
44	Male	KOHPARE UJWAL SUNILRAO	7
46	Male	LIHITKAR ASHITOSH RAVINDRA	5
48	Female	MAHAKULKAR SAKSHI RAJU	2
49	Female	MENDHE VAISHNAVI TATOBA	4
51	Female	MESHRAM RAVINA EKNATH	0
52	Male	MESHRAM YASHWANT PRAKASH	7
53	Female	MILMILE RENUKA SAMBAJI	6
55	Male	NAITAM DHANRAJ PRAKASH	6
56	Male	NANNAWARE SHESHARAO BHARAT	5
59	Female	NIMJE ROSHANI KAVINDRA	4
64	Female	PATHAN MOHEENA PARVIN SAMSHER KHAN	3
66	Female	PIJDURKAR SHRADDHA SAINATH	7
70	Female	RAUT AVANTIKA YOGESH	6
71	Female	SAPAT SAYLI PANDURANG	7
74	Male	SOOR ANKIT GANESH	2
75	Female	TAJNE LAXMI CHANDU	7
79	Male	THOMBARE RAKESH KHUSHALRAO	7
83	Female	WATKAR SAKSHI ANILRAO	4
86	Female	BADKHAL PRATIKSHA VILAS	6
87	Male	BARDE ANIKET GUNWANT	7
89	Male	BHALME MAYUR DILIP	6
90	Male	CHAUDHARI KUNAL SHALIK	7
91	Female	CHINCHOLKAR NIKHITA VITTHAL	6
93	Male	DHAWAS PIYUSH GAJANAN	6
98	Male	JUMADE SANGHARSH RAJHANS	
99	Male	KHIRATKAR OM SANJAY	
100	Female	KUREKAR SHRUTI ANIL	6
101	Male	MAGARE PRAJWAL DAMODHAR	7
105	Male	MESHRAM PRATHAM RAJU	6

106	Female	PIJDURKAR AMBIKA VINAYAK	7
108	Female	SHENDE AMISHA RAMESHRAO	7

SURPRISE TEST ON VECTOR ANALYSIS

SUBJECT TEACHER: ASST. PROF. KALYANI VITTHAL ATRAM

*	Required			
1.	Email *			
2.	NAME *			
3.	MOBILE NUMBER *			
4.	1. The quantity which has the only magnitude	is called	*	2 points
	Mark only one oval.			
	a) A scalar quantity			
	b) A vector quantity			
	c) A chemical quantity			
	d) A magnitude quantity			

5.	2. What is the magnitude of a unit vector? *	2 points
	Mark only one oval.	
	a) It has no magnitude	
	b) Zero	
	c) Constant but not zero	
	d) Unity	
6.	3. Which of the following has zero magnitude? *	2 points
	Mark only one oval.	
	a) Fixed vector	
	b) Zero vector	
	c) Modulus of a vector	
	d) Unit vector	
7.	4. he position of a particle in a rectangular coordinate system is (3,2,5). What will the position vector be? *	2 points
	Mark only one oval.	
	a) 3i^+2j^+5k^	
	b) 3i^-2j^-5k^	
	c) 5i^+2j^+3k^	
	() 2i^+5j^+3k^	

8.	5. Which of the following is a vector? *	2 points
	Mark only one oval.	
	a) Surface Tension	
	b) Moment of Inertia	
	c) Pressure	
	d) None of the above	
0		
9.	6. A quantity described by magnitude and direction both are called *	2 points
	Mark only one oval.	
	a) scalar	
	b) vector	
	c) unit	
	d) all of above	
10.	7. Displacement is *	2 points
	Mark only one oval.	
	a) scalar	
	b) vector	
	c) unit	
	d) all of above	

11.	8. Gradient of scalar field is*	2 points
	Mark only one oval.	
	a) scalar quantity	
	b) vector quantity	
	c) Both a and b	
	d) None of the above	
12.	9. Divergence of vector field is*	2 points
	Mark only one oval.	
	a) scalar quantity	
	b) vector quantity	
	c) Both a and b	
	d) None of the above	
13.	10. Curl of a vector field is*	2 points
	Mark only one oval.	
	a) scalar quantity	
	b) vector quantity	
	c) Both a and b	
	d) None of the above	

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Google Forms

Department of Zoology

Assessment of Slow and Advanced Learner

Name of Student: Class:

Max. Marks: 50 Time: 45mins

Note: All Questions are compulsory and carries 2marks each

- 1. Animals undergo inactive stage during winters. It is called:
 - (a) Acclimatisation (b) Hibernation (c) Aestivation (d) Adaptation
- 2. Ozone protects biosphere from high energetic:
 - (a) Infra-red rays (b) Ultraviolet rays (c) X-rays (d) gamma rays
- 3. Which of the following groups is formed only of the hermaphrodite organisms?
 - (a) Earthworm, tapeworm, housefly, frog
 - (b) Earthworm, tapeworm, sea horse, housefly
 - (c) Earthworm, leech, sponge, roundworm
 - (d) Earthworm, tapeworm, leech, sponge
- 4. Meiosis does not occur in
 - (a) asexually reproducing diploid individuals
 - (b) sexually reproducing haploid individuals
 - (c) sexually reproducing diploid individuals
 - (d) all of these.

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- 5. _____ is a life process that is not essential for an individual's survival but for survival of
 - (a) Growth (b) Reproduction (c) Respiration (d) Nutrition
- 6. Clones' are individuals that have exactly the same
 - (a) Lifespan (b) physiology (c) growth rate (d) genetic makeup.
- 7. Which one of the following processes results in the formation of clone of bacteria?
 - (a) Regeneration (b) Budding (c) Binary fission (d) Fragmentation
- 8. Asexual reproduction is seen in members of Kingdom
 - (a) Monera (b) Plantae (c) Animalia (d) All of these
- 9. During binary fission in Amoeba which of the following organelles is duplicated?
 - (a) Plasma membrane (b) Nucleus (c) Contractile (d) All of these
- 10. Sexual reproduction is considered more beneficial than asexual reproduction because
 - (a) it is not affected by adverse environmental conditions
 - (b) fertilization is a chance factor
 - (c) it rapidly multiplies the population
 - (d) it assists in evolution by producing variations.
- 11. Development of new individual from female gamete without fertilisation is termed as
 - (a) syngamy (b) embryogenesis (c) oogamy (d) parthenogenesis.

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12. Deposition of calcareous shell around zygote occurs in (a) birds and reptiles (b) birds and mammals (c) mammals and reptiles (d) all of these.
13. Select the option which shows viviparous animals only, (a) Lizard, Turtle (b) Platypus, Crocodile (c) Cow, Crocodile (d) Whale, Mouse
14. Viviparity is found in (a) Sharks (b) lizards (c) frogs (d) birds
15. Animal husbandry and plant breeding programmes are the examples of(a) reverse evolution (b) aritifical selection (c) mutation (d) natural selection.
16. Linnaeus evolved a system of nomenclature called (a) Vernacular (b) Monomial (c) Polynomial (d) Binomial
17. What is a taxon? (a) A group of related families (b) A type of living organisms (c) A group of related species (d) A group of any ranking
 18. Five kingdom system of classification suggested by R.H. Whittaker is not based on: (a) Presence or absence of a well defined nucleus. (b) Mode of reproduction. (c) Mode of nutrition. (d) Complexity of body organisation.
19. Who is called as "Father of taxonomy"(a) Hooker (b) Theophrastus (c) Aristotle (d) Linnaeus
20. Sweetest sugar is (a) Fructose (b) Sucrose (c) Glucose (d) Lactose
 21. Abiogenesis theory of origin supports (a) spontaneous generation (b) origin of life from blue-green algae (c) origin of life is due to pre-existing organisms (d) organic evolution is due to chemical reactions.
22. Who proposed that the first form of the could have come from pre-existing non-living
organic molecules? (a) S.L. Miller (b) Oparin and Haldane (c) Charles Darwin (d) Alfred Wallace
23 According to one of the most widely accepted theories, earth's atmosphere before origin of
life was (a) oxidising (b) oxidising along with H_2 (c) reducing with free O_2 in small amount (d) reducing with oxygen absent in O_2 form.
24. According to Oparin, which one of the following was not present in the primitive atmosphere of the earth? (a) Methane (b) Oxygen (c) Hydrogen (d) Water vapour
25. The first life originated (a) on land (b) in air (c) in water (d) all of these.

Professor & Head
P. G. Department of Zoology
A. N. College, Anandwan, Warora

Anand Niketan College, Anandwan, Warora Department of Zoology Slow Learners and Advance Learners List

Sr. No.	Name of the student	Marks Obtained	Remark
1.	Adnan A kha Pathan	38	AL
2.	Nikita Raju Tadas	34	
3.	Sakshi Prakash Majgaoli	32	
4.	Sweta Ishwar Bodbe	12	SL
5.	Tanuja Kawadu Rout	24	
6.	Deepa Narbahadur Thakur	26	
7.	Sakshi Suresh Mohate	34	
8.	Pranali Devanand fulzele	12	, SL
9.	Shreya Vilas Talvekar	24	
10.	Tejaswini Gopal Ganfade	26	
11.	Sakshi Wasudeo Raut	22	
12.	Monica Ulhas Lambat	24	
13.	Diksha Harishchandra Ghugul	26	
14.	Nandini Dnyaneshwar Hepat	40	AL
15.	Gauri Madhukar Titare	36	AL
16.	Komal Sanjay Tadas	32	
17.	Adity L. Parchake	36	AL
18.	Kunal Satpute	38	AL
	Yuvraj Vitthal Lambat	24	7.0
19. 20.	Alok Abhijit Thamke		AL
	Tina Bhaurao Kale	34	7.2
21 .		26	
22	Roshan waman Gajbhe	18	SL
23	Tejaswini V Lodhi	14	SL
24	Harshada D Bawane	42	JL .
25	Vijay Suresh Bodhe	44	
26	Shrushti vijay Jamgade	42	
27	Sayali Raju Bibte	08	SI.
28	Sakshi Arun Bodhe		SL
29	Hemanti S. Bhujade	24	Cl
30	Sneha S. Gawande	18	SL
31	Harsh S. Karwatkar	08	SL
32	Samruddhi Sunil Gawande	22	
33	Nikita dilip dahule	14	SL
34	Gauri Sumersingh Chandel	18	SL
35	Dipti R. Wakde	10	SL
36	Prajakta Kishor Lokhande	18	SL
37 .	Pooja Kailas Parkhi	10	SL
38	Kajal Shankar Rasekar	16	SL
39	Priyanka Ganpat Wandhare	14	SL
40	Vaibhav vitthal tajne	30	
41	Ashwinin raju Gawande	22	
42	Pratiksha Shrikrushna Dethe	16	SL
43	Mayur Umakant Pittalwar	24	
44	Nikita Narayan Mahure	10	SL
45	Divyani Ravindra Thute	10	SL
46	Prerana C. Nanadagade	16	SL
47	Achal Babarao Wankhede	12	SL
48	Prajwal Ankush Ghotkar	08	SL

25

Anand Niketan College, Anandwan, Warora Department of Zoology

Slow Learners and Advance Learners List

	Slow Learners and Advance Learners List				
49	Ajit Vitthal Pajdurkar	18	SL		
50	Sheetal C. Prasad	10	SL		
51	Snehal G. More	14	SL		
52	Aditya Ravindra Chaudhari '	08	SL		
53 .	Arti Janardhan Shende	20			
54	Pratiksha Santosh Thawari	18	SL		
55	Birshti K. Mandal	18	SL		
56	Vaishnavi shankar Nibrad	34			
57	Ashwini Pundalik Nannaware	34			
58	Ashwini Dinkar Chaudhari	28			
59	Gayatri Sanjay Burhan	30			
60	Sanika Ashok Umare	30			
61	Shruti Manohar Lonbale	36	AL		
62	Sakshi Pramod Nande	32			
63	Ankita Dilip Dable	30			
64	Bhagyashri A. Dhongade	32			
65	Achal Gajanan Kashti	12	SL		
66	Prachi Vitthal Ghodmare	36	AL		
67	Sonali Ramesh Datarkar .	26			
68	Jaya Nagoji Harbade	32			
69 .					
70		Mariam	Me		
71			LIERO		
72		Professor	of Zoology		
73		P. G. Departme	andwan, Wall		
74		P. College, A	nt of Zoology		
75		A. N.			
76					
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Department of Zoology

Slow Learners List

Sr. No.	Name of the student	Marks Obtained	Remark
1	Sweta Ishwar Bodbe	12	SL
2	Pranali Devanand fulzele	12	SL
3	Tejaswini V Lodhi	18	SL
4	Harshada D Bawane	14	SL
5	Sakshi Arun Bodhe	08	SL
6	Sneha S. Gawande	18	SL
7	Harsh S. Karwatkar	08	SL
8	Nikita dilip dahule	14	SL
9	Gauri Sumersingh Chandel	18	SL
10	Dipti R. Wakde	10	SL
11	Prajakta Kishor Lokhande	18	SL
12	Pooja Kailas Parkhi	10	SL
13	Kajal Shankar Rasekar	16	SL
14	Priyanka Ganpat Wandhare	14	SL
15	Pratiksha Shrikrushna Dethe	16	SL
16	Nikita Narayan Mahure	10	SL
17	Divyani Ravindra Thute	10	SL
18	Prerana C. Nanadagade	16	SL
19	Achal Babarao Wankhede	12	SL
20	Prajwal Ankush Ghotkar	08	SL
21	Ajit Vitthal Pajdurkar	18	SL
22	Sheetal C. Prasad	10	SL
23	Snehal G. More	14	SL
24	Aditya Ravindra Chaudhari	08	SL
25	Pratiksha Santosh Thawari	18	SL
26	Birshti K. Mandal	18	SL
27	Achal Gajanan Kashti	12	SL

Advance Learners List

Sr. No.	Name of the student	Marks Obtained	Remark
1	Nandini Dnyaneshwar Hepat	40	AL
2	Gauri Madhukar Titare	36	AL
3	Aditya L. Parchake	36	AL
4	Kunal Satpute	38	AL
5	Alok Abhijit Thamke	38	AL
6	Shruti Manohar Lonbale	36	AL
7	Prachi Vitthal Ghodmare	36	AL

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A. N. College, Anandwan,